

CHEMISTRY

FORM FIVE

FIRST TERM – 03.01.2007 to 09.03.2007 (10 weeks)

19.03.2007 to 25.05.2007 (10 weeks)

THEME
LEARNING AREA: INTERACTION BETWEEN CHEMICALS
: 1. RATE OF REACTION

No. of week/ Weeks	Learning Objectives	Suggested Learning Activities	Learning Outcomes	Thinking Skills	Scientific Skills	Scientific Attitudes And Noble Values	Remarks/ Vocabulary
1	1.1 Analysing rate of reaction	Discuss : (a) the meaning of rate of reaction, (b) some examples of fast reactions, (c) some examples of slow reactions. Discuss to identify observable changes to reactants or products and its method of measurement in order to determine the rate of reaction. Carry out an activity involving a reaction between zinc and acid, and plot a graph to determine average rate of reaction and the rate of reaction at any given time. Carry out problem solving activities involving rates of reaction.	A student is able to: - state what rate of reaction is, - identify observable changes to reactants or products for determining rate of reaction, - determine average rate of reaction, - determine the rate of reaction at any given time from a graph, - solve numerical problems involving average rate of reaction, - solve numerical problem involving rate of reaction at any given time.	- generation of ideas - relating making hypothesis - making conclusion - problem solving	- Observing - Hypothesising - Communicating - Interpreting data	- Being honest and accurate in recording and validating data. - Being cooperative - Being systematic	reactant- bahan tindak balas product- hasil tindak balas rate of reaction- kadar tindak balas observable change- perubahan yang dapat diperhatikan

No. of week/ Weeks	Learning Objectives	Suggested Learning Activities	Learning Outcomes	Thinking Skills	Scientific Skills	Scientific Attitudes And Noble Values	Remarks/ Vocabulary
2	1.2 Synthesising factors affecting the rate of reaction	Discuss possible factors affecting the rate of reaction. Design and carry out activities to investigate factors affecting the rate of reaction, i.e. size of reactant, concentration, temperature and catalyst. Some suggested reactions: (a) a reaction between calcium carbonate, CaCO_3, and hydrochloric acid, HCl, (b) a reaction between sodium thiosulphate, $\text{Na}_2\text{S}_2\text{O}_3$, and sulphuric acid, H_2SO_4, (c) decomposition of hydrogen peroxide, H_2O_2, in the presence of a catalyst. Use computer simulations to investigate how the movement and collision of particles in a reaction are affected by temperature, size of reactant, pressure, concentration and catalyst. Collect and interpret data to explain factors affecting the rate of reaction in the following: (a) combustion of charcoal, (b) storing food in a	A student is able to: - design experiment to investigate factors affecting the rate of reaction, - give examples of reactions that are affected by size of reactant, concentration, temperature and catalyst, - explain how each factor affects the rate of reaction, - describe how factors affecting the rate of	- generation of ideas - relating making hypothesis - analyzing - contributing - making conclusion - predicting making inference - problem solving	- experimenting - predicting - interpreting data - inferring	- being honest and accurate in recording and validating data - being systematic - being cooperative - being confident and independent	catalyst - mampin decomposition- penguraian

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