

**Assignment #3 – Compound Names and Formulas**  
**Single and Multi-valent ions**

A. Write the correct formula for the following compounds.

- |                                                    |                                                   |
|----------------------------------------------------|---------------------------------------------------|
| 1. gallium oxide <u><math>Ga_2O_3</math></u>       | 11. silicon oxide <u><math>SiO_2</math></u>       |
| 2. mercury (I) chloride <u><math>HgCl</math></u>   | 12. aluminum bromide <u><math>AlBr_3</math></u>   |
| 3. tin (IV) bromide <u><math>SnBr_4</math></u>     | 13. tin (II) sulphide <u><math>SnS</math></u>     |
| 4. indium oxide <u><math>In_2O_3</math></u>        | 14. germanium bromide <u><math>GeBr_4</math></u>  |
| 5. cadmium sulphide <u><math>CdS</math></u>        | 15. nickel (II) oxide <u><math>NiO</math></u>     |
| 6. magnesium oxide <u><math>MgO</math></u>         | 16. strontium astatide <u><math>SrAt_2</math></u> |
| 7. potassium nitride <u><math>K_3N</math></u>      | 17. aluminum oxide <u><math>Al_2O_3</math></u>    |
| 8. iron (III) sulphide <u><math>Fe_2S_3</math></u> | 18. copper (II) oxide <u><math>CuO</math></u>     |
| 9. gold (III) bromide <u><math>AuBr_3</math></u>   | 19. lawrencium fluoride <u><math>LrF_3</math></u> |
| 10. zinc oxide <u><math>ZnO</math></u>             | 20. actinium oxide <u><math>Ac_2O_3</math></u>    |

B. Name the following compounds correctly. Only include the Roman Numeral in the name if the first element has 2 or more combining capacities.

- |                                        |                                           |
|----------------------------------------|-------------------------------------------|
| 1. $CaBr_2$ <u>Calcium Bromide</u>     | 9. $AsCl_5$ <u>Arsenic (V) Chloride</u>   |
| 2. $K_2O$ <u>Potassium Oxide</u>       | 10. $Eu_2O_3$ <u>Europium (III) Oxide</u> |
| 3. $Fe_2O_3$ <u>Iron (III) Oxide</u>   | 11. $NdBr_3$ <u>Neodymium Bromide</u>     |
| 4. $ZnF_2$ <u>Zinc Fluoride</u>        | 12. $Cu_3N_2$ <u>Copper (II) Nitride</u>  |
| 5. $CuCl$ <u>Copper (I) Chloride</u>   | 13. $CoS$ <u>Cobalt (II) Sulphide</u>     |
| 6. $Ni_2O_3$ <u>Nickel (III) Oxide</u> | 14. $ZrO_2$ <u>Zirconium Oxide</u>        |
| 7. $AuCl$ <u>Gold (I) Chloride</u>     | 15. $Li_2O$ <u>Lithium Oxide</u>          |
| 8. $HgBr$ <u>Mercury (I) Bromide</u>   | 16. $NaH$ <u>Sodium Hydride</u>           |

# Naming Ionic Compounds If8766 Answer

**Michael Seilmaier**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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## **Table of Contents Naming Ionic Compounds If8766 Answer**

1. Understanding the eBook Naming Ionic Compounds If8766 Answer
  - The Rise of Digital Reading Naming Ionic Compounds If8766 Answer
  - Advantages of eBooks Over Traditional Books
2. Identifying Naming Ionic Compounds If8766 Answer
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Naming Ionic Compounds If8766 Answer
  - User-Friendly Interface
4. Exploring eBook Recommendations from Naming Ionic Compounds If8766 Answer
  - Personalized Recommendations
  - Naming Ionic Compounds If8766 Answer User Reviews and Ratings
  - Naming Ionic Compounds If8766 Answer and Bestseller Lists
5. Accessing Naming Ionic Compounds If8766 Answer Free and Paid eBooks
  - Naming Ionic Compounds If8766 Answer Public Domain eBooks
  - Naming Ionic Compounds If8766 Answer eBook Subscription Services
  - Naming Ionic Compounds If8766 Answer Budget-Friendly Options
6. Navigating Naming Ionic Compounds If8766 Answer eBook Formats

- ePub, PDF, MOBI, and More
- Naming Ionic Compounds If8766 Answer Compatibility with Devices
- Naming Ionic Compounds If8766 Answer Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Naming Ionic Compounds If8766 Answer
  - Highlighting and Note-Taking Naming Ionic Compounds If8766 Answer
  - Interactive Elements Naming Ionic Compounds If8766 Answer
- 8. Staying Engaged with Naming Ionic Compounds If8766 Answer
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Naming Ionic Compounds If8766 Answer
- 9. Balancing eBooks and Physical Books Naming Ionic Compounds If8766 Answer
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Naming Ionic Compounds If8766 Answer
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Naming Ionic Compounds If8766 Answer
  - Setting Reading Goals Naming Ionic Compounds If8766 Answer
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Naming Ionic Compounds If8766 Answer
  - Fact-Checking eBook Content of Naming Ionic Compounds If8766 Answer
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
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web description the front of the card is embossed with an image of a ferris wheel alongside london view open up to reveal a three dimensional pop of the iconic london eye one stunning and fascinating landmark in the city grandgift handmade papercraft 3d pop up cards size 18 x 13 cm folded color red default the f

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