

Practical Guide to Naming Oxyacids

- The names for polyatomic ions follow a similar set of rules to those for naming oxyacids, so the name of the polyatomic ion reveals the name of the acid.



Number of oxygens compared to most common	Polyatomic ion			Acid name	
	Example	Prefix	Suffix	Prefix	Suffix
Two Less	ClO^-	hypo	ite	hypo	ous
One Less	ClO_2^-	–	ite	–	ous
Same	ClO_3^-	–	ate	–	ic
One More	ClO_4^-	per	ate	per	ic

- Examples
 - ClO^- is hypochlorite, so HClO is hypochlorous acid.
 - NO_2^- is nitrite, so HNO_2 is nitrous acid.

Naming Acids And Bases Evan P

Kristi Lew



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