

## Naming Ionic Compounds

What are the structural units that make up ionic compounds and how are they named?

Why?

When working in chemistry, it is often convenient to write a chemical in symbols. For example we might write down the substance table salt as  $\text{NaCl}$ . In talking about chemistry however, it is a bit tacky to say "en-ay see-ell" when we want to refer to a substance. Also, in formal writing we should use the name of the compound rather than its symbols. Therefore we need to learn how to say the proper names of ionic substances.

### Model 1 – Ion Charges for Selected Elements

|             |                 |                  |                                      |                                      |                                     |                                                   |  |  |                  |                                      |                                      |                 |                 |                 |  |  |
|-------------|-----------------|------------------|--------------------------------------|--------------------------------------|-------------------------------------|---------------------------------------------------|--|--|------------------|--------------------------------------|--------------------------------------|-----------------|-----------------|-----------------|--|--|
| 1           | H <sup>+</sup>  |                  |                                      |                                      |                                     |                                                   |  |  |                  |                                      |                                      |                 |                 |                 |  |  |
| 2           | Li <sup>+</sup> | Be <sup>2+</sup> | Transition elements                  |                                      |                                     |                                                   |  |  | Al <sup>3+</sup> |                                      | N <sup>3-</sup>                      | O <sup>2-</sup> | F <sup>-</sup>  |                 |  |  |
| 3           | Na <sup>+</sup> | Mg <sup>2+</sup> |                                      |                                      |                                     |                                                   |  |  |                  |                                      | P <sup>3-</sup>                      | S <sup>2-</sup> | Cl <sup>-</sup> |                 |  |  |
| 4           | K <sup>+</sup>  | Ca <sup>2+</sup> | Fe <sup>2+</sup><br>Fe <sup>3+</sup> | Ni <sup>2+</sup><br>Ni <sup>3+</sup> | Cu <sup>+</sup><br>Cu <sup>2+</sup> | Zn <sup>2+</sup>                                  |  |  |                  |                                      |                                      |                 |                 | Br <sup>-</sup> |  |  |
| 5           | Rb <sup>+</sup> | Sr <sup>2+</sup> |                                      |                                      | Ag <sup>1+</sup>                    |                                                   |  |  |                  |                                      | Sn <sup>2+</sup><br>Sn <sup>4+</sup> |                 |                 | I <sup>-</sup>  |  |  |
| 6           |                 | Ba <sup>2+</sup> |                                      |                                      |                                     | Hg <sub>2</sub> <sup>2+</sup><br>Hg <sup>2+</sup> |  |  |                  | Pb <sup>2+</sup><br>Pb <sup>4+</sup> |                                      |                 |                 |                 |  |  |
| ← Cations → |                 |                  |                                      |                                      |                                     |                                                   |  |  |                  |                                      |                                      |                 |                 |                 |  |  |

1. Based on the information in Model 1:

- a. Identify three elements that form only one cation.  $\leftarrow$  Positive Ion  
 $\text{Na}^+$ ,  $\text{Sr}^{2+}$ ,  $\text{Al}^{3+}$  (Any box with one element works.)
- b. Identify three elements that form only one anion.  $\leftarrow$  Negative Ion  
 $\text{P}^{3-}$ ,  $\text{O}^{2-}$ ,  $\text{Br}^-$  (Any box with one element works.)
- c. Identify three elements that form more than one cation.  
 $\text{Fe}^{+2}$  and  $\text{Fe}^{+3}$ ,  $\text{Ni}^{+2}$  and  $\text{Ni}^{+3}$ ,  $\text{Cu}^{+}$  and  $\text{Cu}^{+2}$
- d. In what region of the periodic table are these "multiple ion" elements usually located?  
Transition elements or "d" block

2. Consider the ions of potassium (K) and sulfur (S). Write chemical formulas for all possible ionic compounds involving these ions, using the simplest ratio(s) of potassium (K) and sulfur (S). Keep in mind that the sum of the charges in an ionic compound must equal zero.
- $2K^+$  and  $S^{2-}$  or  $K_2S$
3. Consider the ions of iron (Fe) and sulfur (S). Write chemical formulas for all possible ionic compounds involving these ions, using the simplest ratio(s) of iron (Fe) and sulfur (S). Keep in mind that the sum of the charges in an ionic compound must equal zero.



# Naming Ionic Compounds Packet

**Mohammed Muzibur Rahman**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

## **Naming Ionic Compounds Packet:**

The Top Books of the Year Naming Ionic Compounds Packet The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous compelling novels enthraling the hearts of readers worldwide. Lets delve into the realm of top-selling books, exploring the fascinating narratives that have charmed audiences this year. Naming Ionic Compounds Packet : Colleen Hoover's "It Ends with Us" This touching tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover expertly weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Naming Ionic Compounds Packet : Delia Owens "Where the Crawdads Sing" This evocative coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens weaves a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These popular novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of compelling stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a masterful and thrilling novel that will keep you guessing until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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