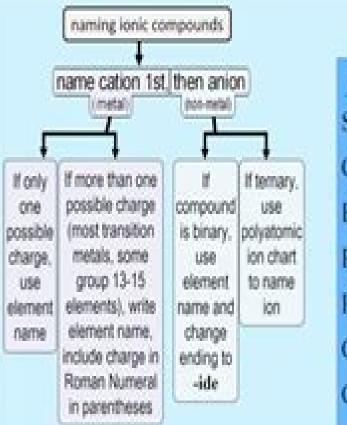
# Naming Ionic Compounds



# Try the following:

FORMULA	NAME
SrCl <sub>2</sub>	strontium chloride
CoCl <sub>2</sub>	cobalt (II) chloride
Ba <sub>3</sub> (AsO <sub>4</sub> ) <sub>2</sub>	barium arsenate
$Pt_3(AsO_4)_2$	platinum (II) arsenate
Pt <sub>3</sub> (AsO <sub>4</sub> ) <sub>4</sub>	platinum (IV) arsenate
Cu <sub>3</sub> N	copper (1) nitride
CuSO <sub>3</sub>	copper (II) sulfite
	1370 1970

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# Name Ionic And Covalent Compounds Guide

Zhe Xu

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