#### Ionic and Covalent Naming Practice Answers

For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately.

1)	Na <sub>2</sub> CO <sub>3</sub>	sodium carbonate	
2)	P4O10	tetraphosphorus decaoxide	
3)	N <sub>2</sub> O <sub>5</sub>	dinitrogen pentoxide	
4)	FeSO <sub>4</sub>	iron (II) sulfate	
5)	SiO <sub>2</sub>	silicon dioxide	
6)	FeCl <sub>3</sub>	iron (III) chloride	
7)	CoBr <sub>2</sub>	cobalt (II) bromide	
8)	B <sub>2</sub> H <sub>4</sub>	diboron tetrahydride	
9)	co	carbon monoxide	
10)	PCIs	phosphorus trichloride	

For each of the following questions, determine whether the compound is ionic or covalent and write the appropriate formula for it.

11)	dinitrogen trioxide	N2O3
12)	nitrogen dioxide	NO <sub>2</sub>
13)	sulfur pentoxide	SOs
14)	lithium acetate	LiC <sub>2</sub> H <sub>3</sub> O <sub>2</sub>
15)	phosphorus trifluoride	PF <sub>3</sub>
16)	vanadium (V) oxide	V <sub>2</sub> O <sub>5</sub>
17)	aluminum hydroxide	AI(OH)a
18)	zinc sulfide	ZnS
19)	silicon tetrafluoride	SiF <sub>4</sub>
20)	silver phosphate	Ag <sub>3</sub> PO <sub>4</sub>

# **Name Or Formulas Of Compounds Answer Key**

**Donald A. McQuarrie, Stanley Gill** 

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#### **Table of Contents Name Or Formulas Of Compounds Answer Key**

- 1. Understanding the eBook Name Or Formulas Of Compounds Answer Key
  - The Rise of Digital Reading Name Or Formulas Of Compounds Answer Key
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Name Or Formulas Of Compounds Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Name Or Formulas Of Compounds Answer Key
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Name Or Formulas Of Compounds Answer Key
  - Personalized Recommendations
  - $\circ\,$  Name Or Formulas Of Compounds Answer Key User Reviews and Ratings
  - Name Or Formulas Of Compounds Answer Key and Bestseller Lists
- 5. Accessing Name Or Formulas Of Compounds Answer Key Free and Paid eBooks
  - Name Or Formulas Of Compounds Answer Key Public Domain eBooks
  - Name Or Formulas Of Compounds Answer Key eBook Subscription Services
  - Name Or Formulas Of Compounds Answer Key Budget-Friendly Options
- 6. Navigating Name Or Formulas Of Compounds Answer Key eBook Formats

- o ePub, PDF, MOBI, and More
- Name Or Formulas Of Compounds Answer Key Compatibility with Devices
- Name Or Formulas Of Compounds Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Name Or Formulas Of Compounds Answer Key
  - Highlighting and Note-Taking Name Or Formulas Of Compounds Answer Key
  - Interactive Elements Name Or Formulas Of Compounds Answer Key
- 8. Staying Engaged with Name Or Formulas Of Compounds Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Name Or Formulas Of Compounds Answer Key
- 9. Balancing eBooks and Physical Books Name Or Formulas Of Compounds Answer Key
  - Benefits of a Digital Library
  - o Creating a Diverse Reading Collection Name Or Formulas Of Compounds Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Name Or Formulas Of Compounds Answer Key
  - Setting Reading Goals Name Or Formulas Of Compounds Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Name Or Formulas Of Compounds Answer Key
  - Fact-Checking eBook Content of Name Or Formulas Of Compounds Answer Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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