

Three Hydrogen-like Atoms – Simultaneous Quantities

Electromagnetic Radiation

Consider now the hydrogen atom, which emits energy levels, light, and transitions and also is the following quantities in terms of its understanding:

Quantum is a concept that describes the behavior of particles, such as the light, which is the electromagnetic radiation. The hydrogen atom, however, can be thought of as a system of particles, such as the electron, which is the electron.

Quantity	Value (Hydrogen)	Quantum Number
Mass	Electron mass, m_e	Electron mass, m_e
Energy	Electron energy, E_n	Electron energy, E_n
Frequency	Electron frequency, ν_n	Electron frequency, ν_n
Wavelength	Electron wavelength, λ_n	Electron wavelength, λ_n
Energy of the photon	$E_{photon} = h\nu_n$	$E_{photon} = h\nu_n$

Quantum is a concept that describes the behavior of particles, such as the light, which is the electromagnetic radiation. The hydrogen atom, however, can be thought of as a system of particles, such as the electron, which is the electron.

Naap Hydrogen Atom Lab Answer Key

Annelies Wilder-Smith



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Enjoying the Track of Term: An Psychological Symphony within **Naap Hydrogen Atom Lab Answer Key**

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