

Name Solutions

Date _____ Period _____

Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
Li	1 ve	transfers $1e^-$ to F	Li^+
F	7 ve	accepts $1e^-$ from Li	F^-

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) $Li + Cl \Rightarrow LiCl$	Li	1 ve	donates $1e^-$	Li^+
$Li + Cl \Rightarrow LiCl$	Cl	1 ve	accepts $1e^-$	Cl^-
2) $Ca + O \Rightarrow CaO$	Ca	2 ve	donates $2e^-$	Ca^{+2}
$Ca + O \Rightarrow CaO$	O	6 ve	accepts $2e^-$	O^{2-}
3) $Be + F \Rightarrow BeF_2$	Be	2 ve	donates $2e^-$ to each F	Be^{+2}
$Be + F \Rightarrow BeF_2$	F	7 ve	each F accepts $1e^-$ each	F^-
4) $Mg + S \Rightarrow MgS$	Mg	2 ve	donates $2e^-$ to S	Mg^{+2}
$Mg + S \Rightarrow MgS$	S	6 ve	accepts $2e^-$	S^{2-}
5) $K + F \Rightarrow KF$	K	1 ve	donates $1e^-$	K^+
$K + F \Rightarrow KF$	F	7 ve	accepts $1e^-$	F^-

Ionic Bonds Review And Reinforce Worksheet Answers

SJ Ball



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Table of Contents Ionic Bonds Review And Reinforce Worksheet Answers

1. Understanding the eBook Ionic Bonds Review And Reinforce Worksheet Answers
 - The Rise of Digital Reading Ionic Bonds Review And Reinforce Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ionic Bonds Review And Reinforce Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ionic Bonds Review And Reinforce Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic Bonds Review And Reinforce Worksheet Answers

- Personalized Recommendations
 - Ionic Bonds Review And Reinforce Worksheet Answers User Reviews and Ratings
 - Ionic Bonds Review And Reinforce Worksheet Answers and Bestseller Lists
5. Accessing Ionic Bonds Review And Reinforce Worksheet Answers Free and Paid eBooks
 - Ionic Bonds Review And Reinforce Worksheet Answers Public Domain eBooks
 - Ionic Bonds Review And Reinforce Worksheet Answers eBook Subscription Services
 - Ionic Bonds Review And Reinforce Worksheet Answers Budget-Friendly Options
 6. Navigating Ionic Bonds Review And Reinforce Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ionic Bonds Review And Reinforce Worksheet Answers Compatibility with Devices
 - Ionic Bonds Review And Reinforce Worksheet Answers Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ionic Bonds Review And Reinforce Worksheet Answers
 - Highlighting and Note-Taking Ionic Bonds Review And Reinforce Worksheet Answers
 - Interactive Elements Ionic Bonds Review And Reinforce Worksheet Answers
 8. Staying Engaged with Ionic Bonds Review And Reinforce Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ionic Bonds Review And Reinforce Worksheet Answers
 9. Balancing eBooks and Physical Books Ionic Bonds Review And Reinforce Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ionic Bonds Review And Reinforce Worksheet Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Ionic Bonds Review And Reinforce Worksheet Answers
 - Setting Reading Goals Ionic Bonds Review And Reinforce Worksheet Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Ionic Bonds Review And Reinforce Worksheet Answers

- Fact-Checking eBook Content of Ionic Bonds Review And Reinforce Worksheet Answers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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