STUDY GUIDE

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write yer if evidence of a chemical reaction is present. Write so if there is no evidence of a chemical reaction.

45

- T. A tomato unella rottes.
- 2. A drinking glass breaks into smaller pieces.

- ma

- 3L. A piece of ior melts.
- 4. Drain cleaner is mixed with water and the solution becomes warm.

465

S. Clandle was burns.

6. Molten caralle was volidifies.

1405

7. Green leaves turn yellow and red as the seasons change.

8. Baking powder produces a gas that makes a coke rise.

In your textbook, read about how to represent chemical reactions and how to belence chemical equations.

Use the terms before to complete the passage. Each term may be used once, more than once, or not at all.

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Guided Study Workbook Chemical Reactions Answers

Sundar Nathan

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Chemical Reactions Science Learning Guide NewPath Learning,2014-03-01 The Chemical Reactions Student Learning Guide includes self directed readings easy to follow illustrated explanations guiding questions inquiry based activities a lab investigation key vocabulary review and assessment review questions along with a post test It covers the following standards aligned concepts Changes of Matter Chemical Reactions Formulas Balancing Equations Types of Chemical Reactions 1 Types of Chemical Reactions 2 Energy in Chemical Reactions Evidence of Chemical Reactions and Chemical Reaction Rates Catalysts Aligned to Next Generation Science Standards NGSS and other state standards

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Swisher, 2023-07-23 NEW Updated content reflects the changes made to the new edition of the Structure and Function text

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