

The Periodic Table of the Elements

The Periodic Table of the Elements

1	2											13	14	15	16	17	18														
1 H Hydrogen												5 B Boron	6 C Carbon	7 N Nitrogen	8 O Oxygen	9 F Fluorine	10 Ne Neon														
3 Li Lithium	4 Be Beryllium											11 Na Sodium	12 Mg Magnesium					13 Al Aluminum	14 Si Silicon	15 P Phosphorus	16 S Sulfur	17 Cl Chlorine	18 Ar Argon								
19 K Potassium	20 Ca Calcium	21 Sc Scandium	22 Ti Titanium	23 V Vanadium	24 Cr Chromium	25 Mn Manganese	26 Fe Iron	27 Co Cobalt	28 Ni Nickel	29 Cu Copper	30 Zn Zinc	31 Ga Gallium	32 Ge Germanium	33 As Arsenic	34 Se Selenium	35 Br Bromine	36 Kr Krypton														
37 Rb Rubidium	38 Sr Strontium	39 Y Yttrium	40 Zr Zirconium	41 Nb Niobium	42 Mo Molybdenum	43 Tc Technetium	44 Ru Ruthenium	45 Rh Rhodium	46 Pd Palladium	47 Ag Silver	48 Cd Cadmium	49 In Indium	50 Sn Tin	51 Sb Antimony	52 Te Tellurium	53 I Iodine	54 Xe Xenon														
55 Cs Cesium	56 Ba Barium	57 La Lanthanum	58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium	71 Lu Lutetium	72 Hf Hafnium	73 Ta Tantalum	74 W Tungsten	75 Re Rhenium	76 Os Osmium	77 Ir Iridium	78 Pt Platinum	79 Au Gold	80 Hg Mercury	81 Tl Thallium	82 Pb Lead	83 Bi Bismuth	84 Po Polonium	85 At Astatine	86 Rn Radon
87 Fr Francium	88 Ra Radium	89 Ac Actinium	90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium	103 Lr Lawrencium	104 Rf Rutherfordium	105 Db Dubnium	106 Sg Seaborgium	107 Bh Bohrium	108 Hs Hassium	109 Mt Meitnerium	110 Ds Darmstadtium	111 Rg Roentgenium	112 Cn Copernicium	113 Nh Nihonium	114 Fl Flerovium	115 Mc Moscovium	116 Lv Livermorium	117 Ts Tennessine	118 Og Oganesson

atomic mass
atomic weight
1st ionization energy

55.845
712.5
7.62

chemical symbol
Fe

name
Iron

electron configuration
(Ar) 3d⁶ 4s²

atomic number
26

electronegativity
1.83

oxidation states
most common and first

alkali metals

alkaline metals

other metals

transition metals

lanthanoids

actinoids

metalloids

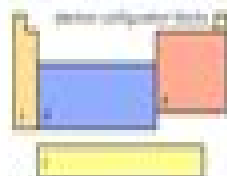
nonmetals

halogens

noble gases

unknown elements

radioactive elements first shown in parentheses



Notes

- 1. Not all elements (113-118) have an official name designation (yet).
- 2. 113-118 are in the bottom row.
- 3. All elements are subject to future name changes.

57 La Lanthanum	58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium
89 Ac Actinium	90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium

Guide To The Periodic Table

Mark Jackson



Guide To The Periodic Table:

Exploring the Elements Isabel Thomas, 2020 Science meets design in this comprehensive introduction to the chemical elements that make up our universe **Your Guide to the Periodic Table**, *The Elements* Albert Stwertka, 2007-03-13 Students of chemistry and of life rejoice Put down your huge textbooks you're free This handy little guide will fit in your back pocket yet give volumes of detail about the elements enough to cover virtually any situation where you'd need to refer to the periodic table This useful tool is an abridged version of the Oxford University Press book *A Guide to the Elements* which was honored with major awards from Science Books and Films magazine and the Society of School Librarians International It also won the Parents Choice Gold Award This tiny edition provides brief surveys of all 114 elements and features an easy to read complete periodic table in a quick foldout The Periodic Table Paul Parsons, 2014-03-11 As one of the most recognizable images in science the periodic table is ingrained in our culture First drawn up in 1869 by Dmitri Mendeleev its 118 elements make up not only everything on our planet but also everything in the entire universe The Periodic Table looks at the fascinating story and surprising uses of each of those elements whether solid liquid or gas From the little known uses of gold in medicine to the development of the hydrogen bomb each entry is accompanied by technical data category atomic number weight boiling point presented in easy to read headers and a colour coding system that helps the reader to navigate through the different groups of elements A remarkable display of thought provoking science and beautiful photography this guide will allow the reader to discover the world afresh The Periodic Table Tom Jackson, 2019 *A Kids' Guide to the Periodic Table* Edward P. Zovinka, Rose A. Clark, 2020-07-21 From aluminum to zinc make it fun for kids 8 to 12 to discover all 118 elements on the periodic table Discover the building blocks of the entire world A Kids Guide to the Periodic Table takes you on an incredible journey through history and science that will teach you all about the 118 elements that make up well everything Go in depth with awesome profiles on each and every element that provide all their important elemental stats like their atomic number state group and more as well as awesome facts about the element and its discovery Take what you know about science and the world to a new level as you discover what makes the periodic table of elements so amazing A Kids Guide to the Periodic Table includes The periodic table explained Learn about the creation of the periodic table and get tons of info to help you understand the groups the order of elements and more Amazing discoveries Explore how elements like neon helium and californium were discovered as well as what they've helped scientists do Fun for you Find out how exciting science can be with an entertaining look into all the ways the elements affect your everyday life A fun fact filled science adventure awaits you with A Kids Guide to the Periodic Table *Your Guide to the Periodic Table* Gill Arbuthnott, 2016-01-04 Your Guide to the Periodic Table is an easy to follow introduction to the elements that make up the periodic table of elements Each element is linked to a science story or fascinating fact from what makes sulfur smelly to what makes hydrogen explode and everything in between Packed with illustrations and explanatory diagrams prepare to be

amazed by the most wacky and informative look at the periodic table ever **A #beginner's Guide to the Periodic Table** Gill Arbuthnott,2014 **Little Book of Elements** Jack Challoner,2020 This little book unlocks each element s colourful history and highlights astonishing facts **Periodic Table (Advanced)** Speedy Publishing,2014-08-01 An advanced periodic table of elements displays not only the elements but the ions that form each element A pamphlet with such a visual aid would greatly benefit chemistry students Any student taking chemistry will need to learn the elements A pamphlet would be concise and break the information down simply making it easier to understand and remember It allows students to simply focus on the main point rather than taking in information that they may or may not need *Chemistry Basics (Speedy Study Guide)* Speedy Publishing,2014-12-28 A study guide of chemistry basics would give students bite sized chunks of information on a complex subject making the task of learning it easier and more enjoyable Study guides present detailed information on a topic in a compact and simplistic way which helps the student grasp more of what they are being taught Also the visual aids that usually are found in study guides can be very effective in keeping the attention of the child as well as embedding the material in their memory *A Beginner's Guide to the Periodic Table* Gill Arbuthnott,2014 A guide to the elements that make up the periodic table fully explaining their starring role in the world and clearing away any confusion or apprehension that might surround them **Elements Explorer** ,1998 **A Guide to the Elements** Albert Stwertka,2012-02-01 Newly updated throughout and now covering 118 elements this crystal clear guide to the periodic table illuminates the basic concepts of chemistry as it traces the history and development of our knowledge of the material world In this fascinating volume Albert Stwertka makes complex ideas and terms easily understandable drawing upon engaging historical anecdotes and everyday examples to clarify the text which is complemented by numerous illustrations many in full color Since the second edition many new elements have been named and discovered including Darmstadtium Roentgenium and Copernicium and the elements currently called Ununtrium Ununpentium Ununhexium Ununseptium and Ununoctium The third edition provides thorough coverage of all these new discoveries In addition to the new elements Stwertka has brought the information about the elements in the second edition up to date based on the latest research He discusses a cylindrical molecule of carbon known as a nanotube which has become a do all wonder substance touted for use in everything from X ray machines to paint A new form of the element boron has been found that is nearly as hard as diamond Its superior heat resistance could make it attractive for certain industrial uses And a new particle detector using ultra pure liquid xenon has been constructed beneath 5 000 feet of rock in Italy to detect dark matter Stwertka also covers the 2010 Nobel winning work on graphene an ultrathin form of carbon that is vital for future generations of computers and touch screens the discovery of new superconductors and the development of new uses for the rare earth elements Bringing the periodic table into the 21st century this engrossing guide to the elements will fascinate everyone curious about the basic building blocks of the material world **My Book of the Elements** Adrian Dingle,2024-03-26 A perfect first book for any young person interested in

chemistry and the world around them My Book of the Elements is a wonderful introduction to the periodic table Covering all the elements from the unreactive to the radioactive as well as key science topics such as states of matter this visual book is something that every young science enthusiast will want to own This series is checked by a specialist consultant It is not only a source of information you can trust but one that is age appropriate and supports your child s schoolwork *HowExpert Guide to Science* HowExpert,2024-11-08 If you re ready to explore the world of science HowExpert Guide to Science is your essential resource covering everything from foundational principles to cutting edge innovations Designed for aspiring scientists students and curious minds this guide takes you on a journey through physics chemistry biology and beyond Focusing on key disciplines interdisciplinary connections ethics and practical knowledge this book offers insights tools and real world examples to truly understand science Chapter Highlights Introduction Discover science s role in everyday life and learn how to get the most from this guide Chapter 1 Foundations of Science Core principles the scientific method and the importance of hypotheses and theories Chapter 2 History of Science Major eras and groundbreaking discoveries that define modern science Chapter 3 Physics Matter energy mechanics thermodynamics and electromagnetism Chapter 4 Chemistry Matter s properties chemical reactions bonding and the periodic table Chapter 5 Biology Genetics evolution ecosystems and cellular biology Chapter 6 Earth Science Geology meteorology oceanography and environmental conservation Chapter 7 Astronomy Our solar system galaxies black holes and cosmology Chapter 8 Human Body Medical Science Human anatomy disease prevention and medical advances Chapter 9 Environmental Science Human impact on the planet sustainability and climate change Chapter 10 Mathematics in Science Math s role in research from probability to calculus Chapter 11 Social Sciences Psychology sociology and the science of human behavior Chapter 12 Technology Applied Sciences Engineering computer science and technological innovation Chapter 13 Data Science Research Research design data analysis and statistical significance Chapter 14 Cutting Edge Sciences Advances in nanotechnology AI quantum mechanics and genetic engineering Chapter 15 Interdisciplinary Science The impact of cross field collaboration and interdisciplinary research Chapter 16 Practical Applications Everyday science in technology health and the environment Chapter 17 Scientific Literacy Skills to evaluate scientific information and avoid misconceptions Chapter 18 Citizen Science How to participate in public science initiatives and projects Chapter 19 Philosophy Ethics of Science Ethical considerations and the role of scientists in society Chapter 20 Science Communication Education Techniques for communicating science and inspiring future generations Chapter 21 Science Careers Explore career paths educational requirements and emerging roles in science Chapter 22 The Future of Science Emerging fields interdisciplinary research and future challenges Conclusion Summarizes key concepts inspires curiosity and reflects on science s impact Appendices Additional resources including a glossary recommended reading notable scientists hands on experiments and science events Why This Guide is Essential for Science Enthusiasts Comprehensive Learning Covers all major branches of science ideal for beginners and enthusiasts alike Practical

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