



Name: _____

Date: _____

Student Exploration: Ionic Bonds

 Vocabulary:
Atom

Use the terminology:

- **Electron**
- **Electron-dot diagram**



Introduction: Some of the earliest chemical experiments were performed using elements that were easily changed, while others were practically unchanged. Elements that appeared changed (1 and 2) and not, while particles with the same charge (3 and 4 or 5 and 6) might

Question: What happens when atoms gain or lose electrons?

1. **Observe** the atoms around the simulation screen. Is oxygen stable, does it tend to gain or lose electrons? How many electrons does it have? How many protons does it have? How many neutrons does it have? How many electrons does it have? How many protons does it have? How many neutrons does it have?

2. Observe the results of the simulation. How many electrons does it have? How many protons does it have? How many neutrons does it have? How many electrons does it have? How many protons does it have? How many neutrons does it have?

Electron: _____ Proton: _____ Neutron: _____

3. How many electrons does it have? How many protons does it have? How many neutrons does it have? How many electrons does it have? How many protons does it have? How many neutrons does it have?

Electron: _____ Proton: _____ Neutron: _____

4. Observe the results of the simulation. How many electrons does it have? How many protons does it have? How many neutrons does it have? How many electrons does it have? How many protons does it have? How many neutrons does it have?

Electron: _____ Proton: _____ Neutron: _____

5. How many electrons does it have? How many protons does it have? How many neutrons does it have? How many electrons does it have? How many protons does it have? How many neutrons does it have?

6. How many electrons does it have? How many protons does it have? How many neutrons does it have? How many electrons does it have? How many protons does it have? How many neutrons does it have?

The number of electrons is _____ The number of protons is _____ The number of neutrons is _____

7. How many electrons does it have? How many protons does it have? How many neutrons does it have? How many electrons does it have? How many protons does it have? How many neutrons does it have?

The number of electrons is _____ The number of protons is _____ The number of neutrons is _____



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R Bogdan

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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