

6 (b) Table 4 shows some properties of gases in the Earth's atmosphere.

Table 4

| Gas            | Melting point in °C | Boiling point in °C |
|----------------|---------------------|---------------------|
| Argon          | -189                | -186                |
| Carbon dioxide | -78                 | -78                 |
| Helium         | -272                | -269                |
| Neon           | -249                | -246                |
| Nitrogen       | -210                | -196                |
| Oxygen         | -219                | -183                |

These gases are separated by:

- removing carbon dioxide
- cooling the remaining gases to  $-200^{\circ}\text{C}$
- removing the gases that do not condense
- allowing the liquefied gases to warm up.

6 (b) (i) Suggest one reason why carbon dioxide is removed before the gases are cooled to  $-200^{\circ}\text{C}$ .

[1 mark]

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6 (b) (ii) Draw a ring around one gas that does not condense when the remaining gases are cooled to  $-200^{\circ}\text{C}$ .

[1 mark]

argon

neon

nitrogen

oxygen

6 (b) (iii) The oxygen separated by this process contains another gas.

Name the gas and give a reason for your answer.

[2 marks]

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