F.Y.B.Sc. Sem II (Regular) Oct 2014 Chemistry II

DURATION: 2 Hrs 30 Min.

Pages 03: Questions 04

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| 1 | All que Figure Diseas | entions a to the next ak | IONS TO THE STUDENTS: are conspalsory. a right indicate the marks. arches / diagrams / graphs / flow sheets wherever necessary. a / scientific calculator / log tables permitted. | |
|----|-----------------------------|--------------------------------|---|----------------------|
| 1. | A) | 90 | Answer <u>Ann. Two</u> of the fell owing. Product the shape of following molecules: BeCl ₂ , NH ₂ , PCl ₃ . | 0.00 |
| | | 600 | g) State and explain foreigntion Energy. q) What is Pajan's Rule? | 03 |
| | | 180 | p) Write short note on : Covalent Bond. q) What are limitations of VSEPB shoory? | 00 |
| | | dey | p) What is Electromogativity? How does it very in period and group in the periodic table? | 00 |
| | | | q) Explain the Heisenberg's principle of Uncertainty? | 00 |
| | 83 | () (0) (80) (80) | Accesver Ann. Two of the fell owing. What is lone pair effect? How does it affect the bond angles in molecules? What are applications of VSEPR theory? Explain: Bf: is planer whereas NH; is not planer. Discuss the Butherlood's atomic model. | 04 04 04 04 |
| 2. | Al | 0 | Attempt Ann Two of the fellowing. p) Define the following - Hydrogen Hend, lente Bond, Dipele Moment. q) Calculate the formal sharge on the atom in the following: Nitrogen in Amenonium ion Oxygen in Water Nitrogen in Nitrogenthane | 03 |
| | | F0.5 | What are Dry Test and Wet Test? Give any two examples for each test. | 1.00 |
| | | 180 | p) What are types of Arrews and their significance? q) Why Formic Acid is stronger than Accid: Acid? | 00 |
| | | 16/3 | Explain the structure of following functional groups: Ether, Cyanide, Ketone. | 00 |
| | mj | 0 10 10 10 10 | Actumpt <u>Ann Two</u> of the following: What is Inductive Effect? Explain its types. What is Senti-micro-Analysis? What is Formal Charge? Explain any two examples. Explain Sp ² hybridization for Nitrogen and Oxygen. | 04 04 04 04 |

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