

Notes on Study-Questions

Changes in Kinetics:

Rate of Reaction:

Rate of Consumption: $-d[A]/dt$

Rate of Change: $d[A]/dt$

Rate of Accumulation: $\text{Accumulation Rate} = \text{Rate of Production} - \text{Rate of Consumption}$

Rate relationships when not:

- isothermal phase changes
- perfect solid/liquid/vapour
- perfect solid/vapour
- distinguish heterogeneous and homogeneous reactions
- isothermal heating/cooling
- ideal & non-ideal solid/liquid/vapour

Ex. 14.24 P. 846

(a) Write an overall balance for the tank

Overall Energy: - the overall heat transfer between the tank and the environment is zero

Enthalpy: - development of energy components in the enthalpy balance eq. 14.24

Enthalpy: - enthalpy change due to phase change

Enthalpy: - the heat transfer

Enthalpy: - the enthalpy change between the inlet and outlet streams

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Enthalpy: - the enthalpy change between the inlet and outlet streams

(b) Write an overall balance for the heat transfer, based on enthalpy

Enthalpy:

(c) Write the overall enthalpy balance and find T_2

Enthalpy:

(d) Write an overall balance for the heat transfer

Enthalpy:

(e) Write an overall balance for the heat transfer, based on enthalpy

Enthalpy:

(f) Write an overall balance for the heat transfer, based on enthalpy

Enthalpy:

(g) Write an overall balance for the heat transfer, based on enthalpy

Enthalpy:

Myers Unit 10 Study Guide Answers

Arturo Cuomo



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