

[8]

[16]

- (v) What is the action of selenium on magnesium metal?
- (vi) Write the name of isomerism in the following complexes: [Cu(NH₃)₄] [PtCl₄] and [Pt(NH₃)₄] [CuCl₄]
- (vii) Write the name of the alloy used in Fischer Tropsch process in the synthesis of gasoline.
- (viii) Henry's law constant for CH₃Br_(g) is 0.159 mol dm⁻³ bar⁻¹ at 25°C. What is solubility of CH₃Br_(g) in water at same temperature and partial pressure of 0.164 bar?

SECTION - B

Attempt any EIGHT of the following questions:

- Q.3. Explain pseudo-first order reaction with suitable example.
- Q.4. Write the consequences of Schottky defect with reasons.
- Q.5. What is the action of following on ethyl bromide:
 - (i) Na in dry ether
 - (iii) Mg in dry ether
- Q.6. Explain formation of peptide linkage in protein with an example.
- Q.7. Derive an expression to calculate molar mass of non volatile solute by osmotic pressure measurement.
- Q.8. Explain monodentate and ambidentate ligands with example.
- Q.9. Explain the trends in the following atomic properties of group 16 elements:

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Table of Contents Hsc Science Question Papers

- 1. Understanding the eBook Hsc Science Question Papers
 - The Rise of Digital Reading Hsc Science Question Papers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Hsc Science Question Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hsc Science Question Papers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Hsc Science Question Papers
 - Personalized Recommendations
 - Hsc Science Question Papers User Reviews and Ratings
 - Hsc Science Question Papers and Bestseller Lists
- 5. Accessing Hsc Science Question Papers Free and Paid eBooks
 - $\circ\,$ Hsc Science Question Papers Public Domain eBooks
 - Hsc Science Question Papers eBook Subscription Services
 - Hsc Science Question Papers Budget-Friendly Options
- 6. Navigating Hsc Science Question Papers eBook Formats

- o ePub, PDF, MOBI, and More
- Hsc Science Question Papers Compatibility with Devices
- Hsc Science Question Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Hsc Science Question Papers
 - Highlighting and Note-Taking Hsc Science Question Papers
 - Interactive Elements Hsc Science Question Papers
- 8. Staying Engaged with Hsc Science Question Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hsc Science Question Papers
- 9. Balancing eBooks and Physical Books Hsc Science Question Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Hsc Science Question Papers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Hsc Science Question Papers
 - Setting Reading Goals Hsc Science Question Papers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Hsc Science Question Papers
 - Fact-Checking eBook Content of Hsc Science Question Papers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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