# Q. 1. Select and write the correct answer for the following multiple choice type of questions :

[10]

The product obtained in the following reaction

 $CH_3 - CH = CH - CH_2 - CHO \xrightarrow{H_2/Ni}$ ? is, (a) Pent-3-en-1-ol (b) Pentan-1-ol

(c) Pentan-2-ol

- (d) Pentanal
- (ii) Amongst the following, the solubility of which ionic solid 030 decreases with increase in temperature?

(b) NaBr

(c) Na,SO,

- (d) KCl
- (iii) The correct IUPAC name of Na<sub>3</sub> [AlF<sub>6</sub>] is

- (a) Sodium hexafluoroaluminate (III)
  - (b) Sodium hexafluoroaluminate (II) (c) Sodium hexafluoroaluminium (III)

  - (d) Sodium hexafluoroaluminium (II)
  - (iv) Which of the following acids has highest pKa value?
    - (a) Mono chloroacetic acid (b) Dichloroacetic acid
    - (c) Trichloroacetic acid (d) Acetic acid
  - (v) Number of carbon atoms present in isoprene unit is \_\_\_\_.

(b) 5

- (d) 3
- (vi) The colourless transition metal ion amongst the following is

(c) Ni<sup>++</sup>

# **Hsc Exam Chemistry Answer Paper**

Lyn Baker

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#### **Table of Contents Hsc Exam Chemistry Answer Paper**

- 1. Understanding the eBook Hsc Exam Chemistry Answer Paper
  - The Rise of Digital Reading Hsc Exam Chemistry Answer Paper
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Hsc Exam Chemistry Answer Paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Hsc Exam Chemistry Answer Paper
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Hsc Exam Chemistry Answer Paper
  - Personalized Recommendations
  - Hsc Exam Chemistry Answer Paper User Reviews and Ratings
  - Hsc Exam Chemistry Answer Paper and Bestseller Lists
- 5. Accessing Hsc Exam Chemistry Answer Paper Free and Paid eBooks
  - Hsc Exam Chemistry Answer Paper Public Domain eBooks
  - Hsc Exam Chemistry Answer Paper eBook Subscription Services
  - Hsc Exam Chemistry Answer Paper Budget-Friendly Options

- 6. Navigating Hsc Exam Chemistry Answer Paper eBook Formats
  - o ePub, PDF, MOBI, and More
  - Hsc Exam Chemistry Answer Paper Compatibility with Devices
  - Hsc Exam Chemistry Answer Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Hsc Exam Chemistry Answer Paper
  - Highlighting and Note-Taking Hsc Exam Chemistry Answer Paper
  - Interactive Elements Hsc Exam Chemistry Answer Paper
- 8. Staying Engaged with Hsc Exam Chemistry Answer Paper
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Hsc Exam Chemistry Answer Paper
- 9. Balancing eBooks and Physical Books Hsc Exam Chemistry Answer Paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Hsc Exam Chemistry Answer Paper
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Hsc Exam Chemistry Answer Paper
  - Setting Reading Goals Hsc Exam Chemistry Answer Paper
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Hsc Exam Chemistry Answer Paper
  - Fact-Checking eBook Content of Hsc Exam Chemistry Answer Paper
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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