

Note:

- All questions are compulsory.
- Answer to the two sections are to be written in the same answer book.
- Figure to the right hand side indicate full marks.
- Write balanced chemical equations and draw neat and labelled diagrams wherever necessary.
- Every new question must be started on a new page.
- Use of logarithmic table is allowed.

CHEMISTRY: SECTION – I

Q. 1. Select and write the most appropriate answer from the given alternatives for each sub-question: [7]

- In body centred cubic structure the space occupied is about
(A) 68 % (B) 53 %
(C) 38 % (D) 32 %
- For a gaseous reaction the unit of rate of reaction is
(A) L atm s^{-1} (B) $\text{atm mol}^{-1} \text{s}^{-1}$
(C) atm s^{-1} (D) mol s
- Which of the following compounds contains S – O as well as S – S bonds?
(A) Sulphuric acid (B) Thiosulphuric acid
(C) Sulphurous acid (D) Thiosulphurous acid
- Which of the following solutions shows maximum depression in freezing point?
(A) 0.5 M Li_2SO_4 (B) 1 M NaCl
(C) 0.5 M $\text{Al}_2(\text{SO}_4)_3$ (D) 0.5 M BaCl_2
- For a chemical reaction $\Delta S = -0.035 \text{ kJ/K}$ and $\Delta H = -20 \text{ kJ}$.
(A) 5.14 K (B) 57.14 K
(C) 571.4 K (D) 5714.0 K
- The standard e.m.f of the following cell is 0.463 V
 $\text{Cu} | \text{Cu}_{(0.01)}^{++} || \text{Ag}_{(0.01)}^{+} | \text{Ag}$. If $E_{\text{Ag}}^{\circ} = 0.800 \text{ V}$,
What is the standard potential of Cu electrode?
(A) 1.137 V (B) 0.337 V
(C) 0.463 V (D) -0.463 V
- Fe_2O_3 is reduced to spongy iron near the top of blast furnace by
(A) H_2 (B) CaO
(C) SiO_2 (D) CO

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Cambridge Checkpoints HSC Chemistry 2017-19 Maureen Slade, Roger Slade, 2016-06-29 **Chemistry**, 2013

Australian Education Index, 1988 *Deep Carbon in Earth: Early Career Scientist Contributions to the Deep Carbon Observatory* Donato Giovannelli, Benjamin A. Black, Alysia D. Cox, Cody S. Sheik, 2017-11-27 Since its inception the Deep Carbon Observatory DCO has coalesced a multidisciplinary and international group of researchers focused on understanding and quantifying Earth's deep carbon budget Carbon is the fourth most abundant element in the universe and understanding carbon chemistry under a variety of environmental conditions impacts all aspects of planetary sciences including planet formation the form and function of planetary interiors and the origin and diversity of life DCO recognizes that is integrating and promoting the contributions of early career scientists are integral to the advancement of knowledge regarding the quantities movements origins and forms of Earth's deep carbon through field experimental analytical and computational research Early career scientists represent the future of deep carbon science and contribute substantially to ongoing research by implementing innovative ideas challenging traditional working schemes and bringing a globally interconnected perspective to the scientific community This research topic highlights the contributions at the forefront of deep carbon research by DCO Early Career Scientist community The manuscripts of this *Frontiers* e volume bear evidence of the rapid advances in deep carbon science and highlights the importance of approaching this field from a plethora of different angles integrating disciplines as diverse as mineralogy geochemistry and microbiology This integration is fundamental in understanding the movements and transformations of carbon across its deep cycle **Lok Sabha Debates** India. Parliament. Lok Sabha, 2012 **Success One HSC Chemistry** Brendan Atkins, Bruce Howarth, Christine Eslick, Leanne Poll, 2013 Success One Chemistry 2018 Edit, Mark Dixon, 2018-02-12 Excel Success One HSC Chemistry contains 2001 2003 and 2008 2016 past HSC questions with detailed answers written by experienced HSC markers a Topic Index a Mark Maximizer Guide and a Glossary of Key Verbs This title helps you get the results you want by practising actual HSC papers and answering HSC level questions *HSC Chemistry* Peter Lewis, Roger Michael Slade, 1986 **HSC Chemistry** Marilyn Schell, Margaret Hogan, 2007-09-01 In this book you will find nearly 600 typical multiple choice examination questions nearly 600 extension questions plus summaries for the Year 12 Chemistry course Answers to all questions are provided Questions follow the dot points in the Board of Studies syllabus and focus on the three core topics plus the option topic Shipwrecks Corrosion and Conservation **Chemistry** Odlum and Garner Staff, 2014 Produced by Science teachers for Science teachers and their students **Creelman HSC Exam Questions** M. Cusiter, V. Cusiter, 2008 *Chemistry*, 2018 Essential Sample HSC Chemistry Papers Timothy Schmidt, David Huang, 1996 **Chemistry**, 2014 Past HSC Papers for Chemistry Includes past HSC papers and worked solutions for the years 2001 to 2014 A comprehensive guide on how to achieve success in the HSC with essential exam techniques and how to study an HSC Examination question by question topic guide glossary

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