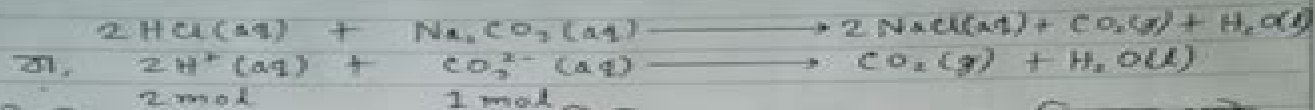


|   |          |    |
|---|----------|----|
| NAME OF THE EXPERIMENT: 0.1M $\text{Na}_2\text{CO}_3$ | Page No. |    |
| উদ্যম দ্বারা সংকলিত নমুনা উদ্যম (HCl/উদ্যম)           | Expt No. | 01 |
| চলমান দিন   | Date :   |    |

মূলনীতি:  $\text{Na}_2\text{CO}_3$  ও  $\text{HCl}$  এর প্রকৃত বিক্রিয়ার সমীকরণ নিম্নরূপ:



টাইট্রেশনের তৈরিতে, প্রকৃত বিক্রিয়া কোথাক প্রকৃত বিক্রিয়ার পরে,  $\text{CO}_3^{2-}$  এর মোল সংখ্যা (ফ্রাক্ট)  $\equiv \text{H}^+$  এর মোল সংখ্যা (ক্যালি) থেকে দেয়া।

প্রয়োজনীয় সামগ্রিক বস্তু:

- ১) 0.1M  $\text{Na}_2\text{CO}_3$  দ্রব
- ২) 0.1M  $\text{HCl}$  দ্রব
- ৩) নির্দেশক : মিথাইল অরেঞ্জ

প্রয়োজনীয় যন্ত্রপাতি:

- ১) কনিক্যাল ফ্লাস্ক
- ২) ব্যালো
- ৩) পিপেট
- ৪) ক্যাম্পাস মেসার

কার্যের ধারা:

- ১) প্রস্তুত করা 0.1M  $\text{HCl}$  দ্রব (প্রকৃত দ্রব) অথবা নমুনা  $\text{HCl}$  দ্বারা ব্যালোটি বর্ণ করে এবং মেসারের মাধ্যমে ক্যাম্পাস দ্বারা পরিমাপ করে আটকিয়ে নিই।
- ২) প্রস্তুত করা 0.1M  $\text{Na}_2\text{CO}_3$  দ্রব (প্রকৃত দ্রব) এর আণবিক ফ্রাক্ট থেকে পিপেট দ্বারা 10ml দ্রব কনিক্যাল ফ্লাস্কে নিই, কনিক্যাল ফ্লাস্কে দ্রব 1-2 ফোটা মিথাইল অরেঞ্জ যোগ করে, তখন দ্রবের বর্ণ হলুদ হয়।
- ৩) কনিক্যাল ফ্লাস্কে তৈরি একটি সাদা কাগজ টুকরা রাখি, ব্যালোটি  $\text{HCl}$  এর দ্রবের দিকে নিয়ে ঘোঁরা দিতে '১ম পাঠ' - এ ফেরত করে, চিহ্নমতে রাস মাতে ব্যালোটির বর্ণ লক্ষ্য করে রাখি। ব্যালোটির  $\text{HCl}$  দ্রব ঘোঁরা ঘোঁরা কনিক্যাল ফ্লাস্কে যোগ করে

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**Ying Liu**



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