

(vi) What is the IUPAC name of



?

(a) α - Amino butanoic acid

(b) 2 - Amino butyric acid

(c) α - Amino butyric acid

(d) 2 - Amino butanoic acid

(vii) Which among the following molecular formulae represents urotropine?

(a) $C_6H_{12}N_4$

(b) $C_6H_{24}H_4$

(c) $C_6H_{12}N_4O_2$

(d) $C_6H_{24}N_4O_2$

Q. 6. Answer any THREE of the following:

[12]

(i) Write the structures of –

(a) 3-chloro-3-ethylhex-1-ene

(b) 1-Iodo-2, 3-dimethylbutane

(c) 1, 3, 5 – tribromobenzene

(ii) What is the action of acidified potassium dichromate on – (a) SO_2 (b) KI

Draw structure of dichromate ion.

(iii) Describe laboratory method for preparation of glucose.

Write the reaction that indicates the presence of –CHO group in glucose.

(iv) What will be the action of the mixture of sodium nitrite and dilute hydrochloric acid on – (a) ethyl amine (b) aniline (c) triethyl amine

Q. 7. Answer any SIX of the following :

[9]

(i) What are chemical twins? Write 'two' examples.

(ii) Explain the terms : (a) Antiseptics (b) Analgesics

(iii) Draw the simple Fisher projection formulae of D – (+) – glucose and D – (–) – fructose.

(iv) Classify the following ligands into monodentate and polydentate –

(a) Ammonia

(b) Carbon monoxide

(c) Ethylene diamine

(d) Ethylene diamine tetra acetate ion

(v) State and explain Markownikoff's rule with suitable example.

(vi) How are propan-1-amine and propan-2-amine prepared from oxime?

(vii) Identify 'A' and 'B' in the following reaction :



(viii) What is the action of the following reagents on phenol –

(a) Bromine in CS_2 at low temperature.

(b) Conc. H_2SO_4 at room temperature.

Q. 8. Answer any ONE of the following:

[7]

(i) Write the structure and IUPAC names of all the metamers represented by formula $C_4H_{10}O$.

Write balanced chemical equations for action of ammonia on –

(a) formaldehyde

(b) acetaldehyde

(c) acetone

(ii) Write 'four' characteristics of co-ordinate complex ion.

How is Nylon 6, 6 prepared?

Write any 'two' uses of terylene.

Write 'four' physical methods of preserving food materials.

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Mike D. Adams



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