এইসএস সি পরীক্ষা ২০১৬ এর মডেল প্রশ্ন (সকল বোর্ডের জন্য)

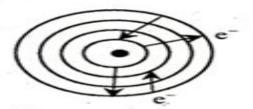
রসায়ন: প্রথম পত্র

সুজনশীল প্রশ্ন

বিষয় কোড : ১ ৭

: সময় — ২ ঘটা

ভিক্তী :- দক্ষিণ পাৰ্কাৰ সংখ্যা প্ৰয়েৱ পূৰ্ণখন জ্ঞাপক। প্ৰদত উন্দীপকালো মনোযোগ দিয়ে পঢ়ো এবং সংক্ৰিট প্ৰৱণুমোৱ কথাকৰ উত্তর দাও। যে কোনো চা রাচি প্রয়েৱ উত্তর দাও। প্রত্যেক প্রয়ের মান ১০ ।/



ক: সাসপেনশন কী?

2. P

খ. কোন্ড ক্রীম গ্রীম্মকালে ব্যবহার করা হয় না কেন?

গ. ইলেকট্রনটি 486.38 nm তরজা বিকিরণ করলে ফোটনের শক্তি কত হবে? ৩

ঘ. উদ্দীপকে প্রদত্ত উচ্চশক্তিস্তর থেকে নিম্নশক্তি স্তরে ঋণাত্মকের ক্ষেত্রে কোনটি অধিক শক্তিশালী তরজা বিকিরণ করে – বিশ্লেষণ কর।

যৌগ	যৌগের মোট ইলেকট্রন
H ₂ X	10
YH.	10
YX2	22

ক. পার আয়োভিক অ্যাসিডের সংকেত লিখ।

খ. সকল অবস্থান্তর মৌলই d-ব্লক মৌল কিন্ত সকল d-ব্লক মৌল অবস্থান্তর মৌল নয়। কেন?

গ. YX2 যৌগটির গঠন অবিটাল সংকরায়ন দ্বারা ব্যাখ্যা কর।

ঘ. "H₂X এবং YH₄ যৌগদ্বয়ের সংকরায়ন অনুরূপ কিন্তু বন্ধন কোণ পরস্পর ভিন্ন।" বিশ্লেষণ কর।

► NA₂ যৌগের পানিতে দ্রাব্যতার গুণফল $4 \times 10^{-4} \left(\frac{\text{mol}}{\text{L}}\right)^3 \text{NA}_2$ যৌগের আণবিক ভর 135.5। NA2 যৌগের একটি সম্পুক্ত দ্রবণে 0.01M ঘনমাত্রার MA যৌগ মিশ্রিত করা হলো। M, N ক্যাটায়ন এবং A অ্যানায়ন। ক. রাইডারের ধ্রবক কী?

খ. ল্যাবরেটরীর নিরাপত্তায় হ্যাজার্ড সিম্বল গুরুত্বপূর্ণ ভূমিকা পালন করে। ব্যাখ্যা কর।

গ. NA, যৌগের দ্রাব্যতা গ্রাম পার লিটার এককে নির্ণয় কর।

বাকী পৃষ্ঠাণ্ডলি নিচের লিঙ্কে ক্লিক করে ডাউনলোড করে পড়ুন।

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