

ক) ভারী হ্যালাজেনের সাথে বিজোড় মণ্ডলিক ইলেকট্রন হ্যালাজেন যুক্ত হয়ে আন্তঃ হ্যালাজেন যৌগ গঠন করে।

৬৬. থ্যালোফেন: থ্যালোফেন নয় কিন্তু থ্যালোফেনের মতোই রাসায়নিক বিক্রিয়া দেয়। যেনন: CN^- , CNS^- ,

ক) ট্রাম-ইউইরেনিয়াম মৌল: (৭৩-১০৩) পারমাণবিক
মধ্যস্থাবিশিষ্ট মৌলগুলোকে ট্রাম-ইউইরেনিয়াম মৌল
বলে।

ক) কৃত্রিম মৌল: (৩৩-১১৪) পারমাণবিক মধ্যখ্যাবিমিশ্র মৌলগুলোকে কৃত্রিম মৌল বলে।

ক) প্রকৃতিতে পাওয়া মৌল ৩০ টি।

উ(১-৩২) পারমানবিক সংখ্যা বিধিষ্ট মৌলগুণের মধ্যে দুটি মৌল প্রকৃতিতে পাওয়া যায় না। তা(৪৩) এক্ষেপম (৬১)।

৯. ১ ব্রক মোলের বহিঃস্থ ইলেকট্রন কাঠামো $\rightarrow ns^{1-2}$

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