

Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 1

Q1. At the same temperature and pressure, an equal volume of gases contains the same number of ____.

- a.) Molecules
- b.) Electrons
- c.) Protons
- d.) Particles

Q2. The gas law that relates the volume of a gas to the number of molecules of the gas is ____.

- a.) Gay-Lussac's law
- b.) Avogadro's law
- c.) Boyle's law
- d.) Charles's law

Q3. The mass % of carbon in CO_2 is ____.

- a.) 0.034%
- b.) 27.27%
- c.) 3.4%
- d.) 28.7%

Q4. The empirical formula of benzene is ____.

- a.) CH
- b.) C_6H_6
- c.) C_3H_3
- d.) C_2H_2

Q5. What is the mole fraction of the solute in a 1.00 molal aqueous solution?

- a.) 0.00177
- b.) 0.0344
- c.) 0.0177
- d.) 0.1770

Q6. State the number of significant figures in each of the following numbers:

- i.) 207.35

Hrw Chemistry Ch 1

G Psacharopoulos



Hrw Chemistry Ch 1:

Hrw Chemistry Ch 1 Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Hrw Chemistry Ch 1**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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