

Edexcel iGCSE Chemistry 4CH0 Learning Plan

Section A2: Atomic structure		
Specification	Resources	Assessment
<p>1.8 recall that atoms consist of a central nucleus, composed of protons and neutrons, surrounded by electrons, orbiting in shells</p> <p>1.9 recall the relative mass and relative charge of a proton, neutron and electron</p> <p>1.10 understand the terms atomic number, mass number, isotopes and relative atomic mass (A_r)</p> <p>1.11 calculate the relative atomic mass of an element from the relative abundances of its isotopes</p> <p>1.12 understand that the Periodic Table is an arrangement of elements in order of atomic number</p> <p>1.13 deduce the electronic configurations of the first 20 elements from their positions in the Periodic Table</p> <p>1.14 deduce the number of outer electrons in a main group element from its position in the Periodic Table.</p>	<p>Video: Section 1 – Lesson 1 – States of Matter and Atomic Structure (from 18:55 to the end).</p> <p>Powerpoint: Chemistry - Section 1 – Lesson 1 (slide 55 to end)</p> <p>Textbook: Page 6 – Chapter 2: Atomic Structure Page 7 – The structure of the atom Page 7 – Atomic number and mass number Page 8 – The electrons; to Page 11</p> <p>Page 12 – Chapter check list</p>	<p>Textbook Page 12 – end of chapter questions.</p> <p>Talking paper – Edexcel Chemistry Section A2 Atomic structure.</p> <p>Section A2 Question – Atomic structure. (pdf)</p> <p>Section A2 Question – mark scheme. (pdf)</p>

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1. Understanding the eBook Gcse Chemistry 2012 Jac 1c
 - The Rise of Digital Reading Gcse Chemistry 2012 Jac 1c
 - Advantages of eBooks Over Traditional Books
2. Identifying Gcse Chemistry 2012 Jac 1c
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gcse Chemistry 2012 Jac 1c
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gcse Chemistry 2012 Jac 1c
 - Personalized Recommendations
 - Gcse Chemistry 2012 Jac 1c User Reviews and Ratings
 - Gcse Chemistry 2012 Jac 1c and Bestseller Lists
5. Accessing Gcse Chemistry 2012 Jac 1c Free and Paid eBooks
 - Gcse Chemistry 2012 Jac 1c Public Domain eBooks
 - Gcse Chemistry 2012 Jac 1c eBook Subscription Services
 - Gcse Chemistry 2012 Jac 1c Budget-Friendly Options
6. Navigating Gcse Chemistry 2012 Jac 1c eBook Formats

- ePub, PDF, MOBI, and More
 - Gcse Chemistry 2012 Jac 1c Compatibility with Devices
 - Gcse Chemistry 2012 Jac 1c Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gcse Chemistry 2012 Jac 1c
 - Highlighting and Note-Taking Gcse Chemistry 2012 Jac 1c
 - Interactive Elements Gcse Chemistry 2012 Jac 1c
 8. Staying Engaged with Gcse Chemistry 2012 Jac 1c
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gcse Chemistry 2012 Jac 1c
 9. Balancing eBooks and Physical Books Gcse Chemistry 2012 Jac 1c
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gcse Chemistry 2012 Jac 1c
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Gcse Chemistry 2012 Jac 1c
 - Setting Reading Goals Gcse Chemistry 2012 Jac 1c
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Gcse Chemistry 2012 Jac 1c
 - Fact-Checking eBook Content of Gcse Chemistry 2012 Jac 1c
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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