

### D. Presence of a Catalyst

- What effect does the  $\text{MnO}_2$  catalyst have on the rate of evolution of  $\text{O}_2$  gas?
- Write a balanced equation for the decomposition of  $\text{H}_2\text{O}_2$ .

### E. Concentration of Reactants: Magnesium-Hydrochloric Acid System

Concentration of HCl	Volume of HCl	mol HCl	mass of Mg	mol Mg	mol HCl / mol Mg	Time (sec)
6 M	5 mL		0.0301			
4 M	5 mL		0.0188			
3 M	5 mL		0.0161			
1 M	5 mL		0.0124			

- Plot  $\frac{\text{mol HCl}}{\text{mol Mg}}$  (y-axis) versus time (x-axis). Instructor's approval of graph: \_\_\_\_\_
- From your graph, predict the time, in seconds, for 5 mg of Mg to react in 5 mL of 2.0 M HCl.

### F. Concentration of Reactants: Iodic Acid-Sulfurous Acid System

Molar concentration of  $\text{HIO}_3$  \_\_\_\_\_ Molar concentration of  $\text{H}_2\text{SO}_3$  \_\_\_\_\_

- What is the volume (mL) per drop of the  $\text{HIO}_3$  solution? \_\_\_\_\_ What is the volume (mL) per drop of the water? \_\_\_\_\_

Test Tube	Drops of $\text{HIO}_3$	mL $\text{HIO}_3$	Drops of $\text{H}_2\text{O}$	mL $\text{H}_2\text{O}$	mL $\text{H}_2\text{SO}_3$	$[\text{HIO}_3]_0$ (diluted)	Time (sec)
1					1.0		
2					1.0		
3					1.0		
4					1.0		
5					1.0		

- Plot  $[\text{HIO}_3]_0$  (y-axis) versus time (x-axis). Instructor's approval of graph: \_\_\_\_\_
- How does a change in the molar concentration of  $\text{HIO}_3$  affect the time required for the appearance of the deep-blue  $\text{I}_2$ -starch complex?
- Estimate the time, in seconds, for the deep-blue  $\text{I}_2$ -starch complex to form when 10 drops of 0.01 M  $\text{HIO}_3$  are used for the reaction. Assume all other conditions remain constant.

<sup>a</sup>See footnote 5.

# Factors Affecting Reaction Rate Lab Answers

**United States. Energy Research and  
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**Laboratory Manual for Principles of General Chemistry** Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures *Laboratory Manual for Principles of General Chemistry* J. A. Beran, Mark Lassiter, 2022-08-16 Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write textbooks textbooks don't write experiments A student's scientific literacy grows when experiences and observations associated with the scientific method are encountered Further experimentation provides additional cause effect observations leading to an even better understanding of the experiment The 11th edition's experiments are informative and challenging while offering a solid foundation for technique safety and experimental procedure The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit An Additional Notes column is included in each experiment's Report Sheet to provide a space for recording observations and data during the experiment Continued emphasis on handling data is supported by the Data Analysis section EPA Publications Bibliography Quarterly Abstract Bulletin United States. Environmental Protection Agency, 1995 **EPA Publications Bibliography** United States. Environmental Protection Agency, 1985 **Subject Index to Unclassified ASTIA Documents** Defense Documentation Center (U.S.), 1960 **Nuclear Science Abstracts**, 1968-07 **National Library of Medicine Current Catalog** National Library of Medicine (U.S.), 1968 **Chemistry** Lemay, Prentice-Hall Staff, 1996 **Report Announcement Bulletin, Unclassified Reports for Civilian Applications** U.S. Atomic Energy Commission, 1956 The U S Atomic Energy Commission is conducting a large scale review of its research and development reports to make as much information as possible available through the Civilian Application Program Report Announcement Bulletin Unclassified Reports For Civilian Applications is being published to announce immediately the release of newly declassified reports All reports announced in the Bulletin are available from Office of Technical Services Department of Commerce Washington 25 D C at the price listed with each title P iii **Scientific and Technical Aerospace Reports**, 1989 *Laboratory Manual* Jo A. Beran, 1990 **Fossil Energy Update**, 1984 Proceedings of the Human Factors Society 28th Annual Meeting Human Factors Society. Meeting, 1984

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In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Factors Affecting Reaction Rate Lab Answers**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound affect on the souls of its readers.

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