

PHYSICAL SCIENCES P.2

MEMO-2024



basic education

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Basic Education
REPUBLIC OF SOUTH AFRICA

G.12

NATIONAL
SENIOR CERTIFICATE

Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



2.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

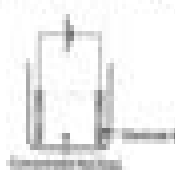
The inorganic reagents shown below are available for reactions IV and Y.

Br ₂	H ₂ SO ₄ (conc.)	NaOH(aq.)	HBr	H ₂
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4.5 Write down the TYPE of elimination in reaction III.

4.6 Compounds R and T are positional isomers.

4.7 Identify the reaction in the above diagram that is a redox reaction.



4.8 Write the half-cell reaction for the anode and the cathode in the above cell.



4.9 Identify the reaction in the above diagram.

Write the half-cell reaction for the anode and the cathode in the above cell.

4.10 Write the half-cell reaction for the anode and the cathode in the above cell.

4.11 The simplified diagram below represents an electrochemical cell. Write the half-cell reaction for the anode and the cathode in the above cell.

Mecanotechinology Memo November 2014

Emilie Sanchez



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