

নিম্নের প্রশ্নসমূহে : সমসাময়িকভাবে অনুবিভাগীয় অধীক্ষকের উপস্থিতিতে প্রাচীরে একটির সমস্ত প্রশ্নের বিশেষায়িত প্রশ্নের অনুবিভাগীয় অধ্যয়ন করে। যে কোনো প্রশ্নেরই প্রাচীরে উত্তর দিতে হবে। : প্রতিটি প্রশ্নের মান ২ (২)

প্রশ্নসমূহের কোনো প্রশ্নের মান/মার্ক দেয়া হবে না।

১. নিম্নের কোনটি সত্যের প্রকাশ?

- (ক) ১০ mL ০.১M NH_4OH + ১০ mL ০.১M HCl
 (খ) ১০ mL ০.১M CH_3COOH + ১০ mL ০.১M NaOH
 (গ) ১০ mL ০.১M NaOH + ১০ mL ০.১M H_2CO_3
 (ঘ) ১০ mL ০.১M NaOH + ১০ mL ০.১M HCl

২. অক্সিজেনের উদাহরণ কোনটি?

- (ক) $^{16}_8\text{O}$
 (খ) $^{14}_6\text{C}$
 (গ) $^{11}_5\text{B}$
 (ঘ) $^{12}_6\text{C}$

৩. নিম্নের কোনটি সত্যের প্রকাশ?

- I. অক্সিজেনের পরিবেশ স্থিতির মাধ্যমে
 II. ব্যাকটেরিয়ার আকর্ষণ সাইট নষ্ট করার মাধ্যমে
 III. কিলোসি, একোসি হিসেবে

নিম্নের কোনটি সত্য?

- (ক) I ও II
 (খ) II ও III
 (গ) I ও III
 (ঘ) I, II ও III

৪. ২৫°C সে. তাপমাত্রায় ১৫০ গ্রাম সম্পৃক্ত দ্রবণে ৫০ গ্রাম দ্রব প্রবর্তিত থাকলে ঐ দ্রবের প্রত্যেক কণা?

- (ক) ১০০
 (খ) ৭৫
 (গ) ৫০
 (ঘ) ২৫

৫. $2\text{SO}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{SO}_3(\text{g}) + 44.8\text{KCal}$

বিক্রিয়াক্রমে তাপমাত্রা বৃদ্ধি করলে—

- I. SO_2 এর পরিমাণ হ্রাস পায়
 II. K_c এর মান হ্রাস পায়
 III. বিক্রিয়া পটভাসুখী হয়

নিম্নের কোনটি সত্য?

- (ক) I ও II
 (খ) II ও III
 (গ) I ও III
 (ঘ) I, II ও III

৬. কোনটির সমস্ত মান $n = 4$ এবং $l = 3$ হলে অক্সিজেনটি হবে—

- (ক) ৬০
 (খ) ৬৫
 (গ) ৬৮
 (ঘ) ৬৯

৭. নিম্নের কোন উপস্থাপন দ্বারা খাদ্য সংরক্ষণ পদ্ধতিতে বিক্রিয়াক্রম?

- (ক) অক্সিজেন কোল
 (খ) গ্লিসি
 (গ) বাল্য শাল
 (ঘ) নিম্নের

৮. পৃষ্ঠ অক্সিজেনের μM ঘনতবে ১.০ ও ৫.০ হলে কোন দ্রবণটি বিক্রিয়াক্রম অপেক্ষা কম ওল বেশি অক্সিজেন?

- (ক) ২০ ওল
 (খ) ১০ ওল
 (গ) ১০০ ওল
 (ঘ) ১০০০ ওল

৯. M^{3+} আয়নে ২৩টি ইলেকট্রন থাকলে M এর পরমাণবিক সংখ্যা কত?

- (ক) ২৩
 (খ) ২৪
 (গ) ২৫
 (ঘ) ২৬

১০. অক্সিজেন বৌল কোনটি?

- (ক) TiCl_4
 (খ) MnO_2
 (গ) ScCl_3
 (ঘ) ZnSO_4

১১. অসংজ্ঞায়িত দ্রবণের সূত্রের পার্থক্যিক প্রকাশ কোনটি?

- (ক) $\alpha = \sqrt{\frac{K_c}{C}}$
 (খ) $\alpha = \sqrt{\frac{C}{K_c}}$
 (গ) $\alpha = \sqrt{\frac{1}{K_c}}$
 (ঘ) $\alpha = \frac{1}{\sqrt{C}}$

(পর পৃষ্ঠা প্রদর্শন)

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