

UNIT III

TUTORIAL 10: K_{sp} CALCULATIONS

Solubility Product Constants (at 25°C)

Substance	K_{sp}	Substance	K_{sp}	Substance	K_{sp}
AgBr	7.70×10^{-13}	BaSO ₄	1.08×10^{-10}	MnCO ₃	1.82×10^{-11}
AgBrO ₃	5.77×10^{-5}	CaCO ₃	8.70×10^{-9}	NiCO ₃	6.61×10^{-9}
Ag ₂ CO ₃	6.15×10^{-12}	CdS	3.60×10^{-29}	PbCl ₂	1.62×10^{-5}
AgCl	1.56×10^{-10}	Cu(IO ₃) ₂	1.40×10^{-7}	PbI ₂	1.39×10^{-8}
Ag ₂ CrO ₄	9.00×10^{-12}	CuC ₂ O ₄	2.87×10^{-8}	Pb(IO ₃) ₂	2.60×10^{-13}
Ag ₂ Cr ₂ O ₇	2.00×10^{-7}	FeC ₂ O ₄	2.10×10^{-7}	SrCO ₃	1.60×10^{-9}
AgI	1.50×10^{-16}	FeS	3.70×10^{-19}	TiBr	3.39×10^{-6}
AgSCN	1.16×10^{-12}	Hg ₂ SO ₄	7.41×10^{-7}	ZnCO ₃	1.45×10^{-11}
Al(OH) ₃	1.26×10^{-33}	Li ₂ CO ₃	1.70×10^{-2}	ZnS	1.20×10^{-23}
BaCO ₃	8.10×10^{-8}	MgCO ₃	2.60×10^{-5}		

Ksp Manual Calculations

Emilie Sanchez



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Table of Contents Ksp Manual Calculations

1. Understanding the eBook Ksp Manual Calculations
 - The Rise of Digital Reading Ksp Manual Calculations
 - Advantages of eBooks Over Traditional Books
2. Identifying Ksp Manual Calculations
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ksp Manual Calculations
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ksp Manual Calculations
 - Personalized Recommendations
 - Ksp Manual Calculations User Reviews and Ratings
 - Ksp Manual Calculations and Bestseller Lists
5. Accessing Ksp Manual Calculations Free and Paid eBooks
 - Ksp Manual Calculations Public Domain eBooks
 - Ksp Manual Calculations eBook Subscription Services
 - Ksp Manual Calculations Budget-Friendly Options
6. Navigating Ksp Manual Calculations eBook Formats

- ePub, PDF, MOBI, and More
- Ksp Manual Calculations Compatibility with Devices
- Ksp Manual Calculations Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ksp Manual Calculations
 - Highlighting and Note-Taking Ksp Manual Calculations
 - Interactive Elements Ksp Manual Calculations
- 8. Staying Engaged with Ksp Manual Calculations
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ksp Manual Calculations
- 9. Balancing eBooks and Physical Books Ksp Manual Calculations
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ksp Manual Calculations
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ksp Manual Calculations
 - Setting Reading Goals Ksp Manual Calculations
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ksp Manual Calculations
 - Fact-Checking eBook Content of Ksp Manual Calculations
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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