

16. The no. of Neutrons in ^{14}N d) 9
 a) 6 b) 7 c) 8
17. The energy of one quantum of radiation is calculated by d) $E = h\nu$
 a) $E = h\nu$ b) $h = E\nu$ c) $h = \nu/E$

(08)

Q.2) Answer the followings.

- What are Physical classification of Matter?
- Define the elements and explains?
- Define Atomic Mass?
- Write the chemical equation for combustion of propane?
- Write down the IUPAC names?
 $\text{CH}_3\text{-C}(\text{C}_2\text{H}_5)_2\text{-CH}_3$ $\text{C}(\text{CH}_3)_4$ -CH_3
- Define the law of conservation of mass?
- If an element X has mass no. 11 and it has 6 neutrons then write its representation.
- State Heisenberg Uncertainty Principle?

Section- B

(16)

Attempt any Eight

- Q.3) How will you convert Propane into @-nitro propane?
- Q.4) How will you convert ethane into ethyl chloride?
- Q.5) State the uses of alkane?
- Q.6) write the physical properties of alkanes?
- Q.7) What are alkane? how they are classified?
- Q.8) Find the no. of Moles (n) present in 72 grams of H_2O ?
- Q.9) Write down the Units of Measurements?
- Q.10) Differentiates between Isotopes and Isobars?
- Q.11) State and Explains Pauli's exclusion principle
- Q.12) State Hund's rule of maximum Multiplicity?
- Q.13) State the order of fillings Atomic Orbital's model

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Table of Contents Jssneco 2examination Question Paper

1. Understanding the eBook Jssneco 2examination Question Paper
 - The Rise of Digital Reading Jssneco 2examination Question Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Jssneco 2examination Question Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Jssneco 2examination Question Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Jssneco 2examination Question Paper
 - Personalized Recommendations
 - Jssneco 2examination Question Paper User Reviews and Ratings
 - Jssneco 2examination Question Paper and Bestseller Lists
5. Accessing Jssneco 2examination Question Paper Free and Paid eBooks
 - Jssneco 2examination Question Paper Public Domain eBooks
 - Jssneco 2examination Question Paper eBook Subscription Services
 - Jssneco 2examination Question Paper Budget-Friendly Options

6. Navigating Jssneco 2examination Question Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - Jssneco 2examination Question Paper Compatibility with Devices
 - Jssneco 2examination Question Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Jssneco 2examination Question Paper
 - Highlighting and Note-Taking Jssneco 2examination Question Paper
 - Interactive Elements Jssneco 2examination Question Paper
8. Staying Engaged with Jssneco 2examination Question Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Jssneco 2examination Question Paper
9. Balancing eBooks and Physical Books Jssneco 2examination Question Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Jssneco 2examination Question Paper
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Jssneco 2examination Question Paper
 - Setting Reading Goals Jssneco 2examination Question Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Jssneco 2examination Question Paper
 - Fact-Checking eBook Content of Jssneco 2examination Question Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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