

**JEE 2009
Paper I**

PART I : CHEMISTRY

SECTION - I

Single Correct Choice Type

This section contains 8 multiple choice questions. Each question has 4 choices (A), (B), (C) and (D) for its answer, out of which **ONLY ONE** is correct.

1. Given that the abundances of isotopes ^{54}Fe , ^{56}Fe and ^{58}Fe are 5%, 90% and 5%, respectively, the atomic mass of Fe is
(A) 55.85 (B) 55.95 (C) 55.75 (D) 56.05

Answer

- (A) ☐ (B) ☒ (C) ☐ (D) ☐

2. The term that corrects for the attractive forces present in a real gas in the van der Waals equation is

- (A) ab (B) $\frac{an^2}{V^2}$ (C) $-\frac{an^2}{V^2}$ (D) $-ab$

Answer

- (A) ☐ (B) ☒ (C) ☐ (D) ☐

3. Among the electrolytes Na_2SO_4 , CaCl_2 , $\text{Al}_2(\text{SO}_4)_3$ and NH_4Cl , the most effective coagulating agent for Sb_2S_3 sol is

- (A) Na_2SO_4 (B) CaCl_2 (C) $\text{Al}_2(\text{SO}_4)_3$ (D) NH_4Cl

Answer

- (A) ☐ (B) ☐ (C) ☒ (D) ☐

4. The Henry's law constant for the solubility of N_2 gas in water at 298 K is 1.0×10^5 atm. The mole fraction of N_2 in air is 0.8. The number of moles of N_2 from air dissolved in 10 moles of water at 298 K and 5 atm pressure is

- (A) 4.0×10^{-4} (B) 4.0×10^{-5} (C) 5.0×10^{-4} (D) 4.0×10^{-6}

Answer

- (A) ☒ (B) ☐ (C) ☐ (D) ☐

5. The reaction of P_4 with X leads selectively to P_4O_6 . The X is

- (A) Dry O_2 (B) A mixture of O_2 and N_2
(C) Moist O_2 (D) O_2 in the presence of aqueous NaOH

Answer

- (A) ☐ (B) ☒ (C) ☐ (D) ☐

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