

Paper 1

Questions are applicable for both core and extended candidates

- 1 Which statement describes methane?
- A It is an alcohol.
 - B It is an unsaturated molecule.
 - C It contains carbon, hydrogen and oxygen atoms only.
 - D Its molecules contain four single covalent bonds.
- 2 Which equation represents the cracking of an alkane?
- A $3\text{C}_2\text{H}_6 \rightarrow \text{C}_6\text{H}_{12}$
 - B $\text{C}_6\text{H}_{12} + \text{H}_2 \rightarrow \text{C}_6\text{H}_{14}$
 - C $\text{C}_6\text{H}_{14} \rightarrow 6\text{C} + 7\text{H}_2$
 - D $\text{C}_6\text{H}_{14} \rightarrow \text{C}_2\text{H}_6 + \text{C}_4\text{H}_{10}$
- 3 Which statement about ethane is correct?
- A It decolourises bromine water.
 - B It burns in excess oxygen to form water and carbon dioxide.
 - C Its molecular formula is C_2H_4 .
 - D Its atoms are joined together by ionic bonding.
- 4 Some properties of colourless liquid L are listed.
- It boils at 65°C .
 - When added to water, two layers form which do not mix.
 - It does not react with sodium carbonate.
 - It has no effect on bromine water.

What is L?

- A ethanol
- B hexane
- C hexene
- D ethanoic acid

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1. Understanding the eBook Igcse Chemistry Examination Paper Jan 24
 - The Rise of Digital Reading Igcse Chemistry Examination Paper Jan 24
 - Advantages of eBooks Over Traditional Books
2. Identifying Igcse Chemistry Examination Paper Jan 24
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Igcse Chemistry Examination Paper Jan 24
 - User-Friendly Interface
4. Exploring eBook Recommendations from Igcse Chemistry Examination Paper Jan 24
 - Personalized Recommendations
 - Igcse Chemistry Examination Paper Jan 24 User Reviews and Ratings
 - Igcse Chemistry Examination Paper Jan 24 and Bestseller Lists
5. Accessing Igcse Chemistry Examination Paper Jan 24 Free and Paid eBooks
 - Igcse Chemistry Examination Paper Jan 24 Public Domain eBooks
 - Igcse Chemistry Examination Paper Jan 24 eBook Subscription Services
 - Igcse Chemistry Examination Paper Jan 24 Budget-Friendly Options
6. Navigating Igcse Chemistry Examination Paper Jan 24 eBook Formats

- ePub, PDF, MOBI, and More
 - Igcse Chemistry Examination Paper Jan 24 Compatibility with Devices
 - Igcse Chemistry Examination Paper Jan 24 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Igcse Chemistry Examination Paper Jan 24
 - Highlighting and Note-Taking Igcse Chemistry Examination Paper Jan 24
 - Interactive Elements Igcse Chemistry Examination Paper Jan 24
 8. Staying Engaged with Igcse Chemistry Examination Paper Jan 24
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Igcse Chemistry Examination Paper Jan 24
 9. Balancing eBooks and Physical Books Igcse Chemistry Examination Paper Jan 24
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Igcse Chemistry Examination Paper Jan 24
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Igcse Chemistry Examination Paper Jan 24
 - Setting Reading Goals Igcse Chemistry Examination Paper Jan 24
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Igcse Chemistry Examination Paper Jan 24
 - Fact-Checking eBook Content of Igcse Chemistry Examination Paper Jan 24
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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