

**BOARD OF INTERMEDIATE EDUCATION, KARACHI****INTERMEDIATE EXAMINATION, 2021 (ANNUAL)****CHEMISTRY PAPER – II****(Science Pre-Engineering & Pre-Medical Groups)**

Max. Marks: 85

Time: 2 hours

**SECTION 'A'****(MULTIPLE CHOICE QUESTIONS) – (M.C.Q.s)****(Marks: 43)****NOTE:**

- i) This section consists of 43 part questions and all are to be answered. Each question carries one mark.  
 ii) Do not copy the part questions in your answerscript. Write only the answer in full against the proper number of the question and its part.

The correct answers are highlighted in red colour.

1. Select the correct answer for each from the given options:

- i) Which pair of elements belongs to same group in the periodic table:  
 \* Mg and S \* O and Br \* Na and Al \* **C and Si**
- ii) Hydride ion and Helium atom have the same:  
 \* Number of protons \* Number of neutrons  
 \* Valency \* **Number of electrons**
- iii) The valence shell electronic configuration of Halogen family is:  
 \*  $ns^1$  \*  $ns^2, np^5$  \*  $ns^2, (n-1)d^5$  \*  **$ns^2, np^5$**
- iv) Dolomite is a carbonate of:  
 \* Ca and Ba \* Li and Na \* Mg and Al \* **Mg and Ca**
- v) Nitric acid is frequently transported in a container made of:  
 \* **Al** \* Cu \* Zn \* Ag
- vi) The mixture of Aluminium powder and Aluminium nitrate is called:  
 \* Gun powder \* **Ammonal** \* Urea \* Tincal
- vii) 'A' is the element of IIA group while 'B' belongs to VA group. When 'A' reacts with 'B' they form:  
 \* AB \*  $AB_3$  \*  **$A_3B_2$**  \*  $A_2B_3$
- viii) Diaspore is an ore of:  
 \* Copper \* Zinc \* Iron \* **Aluminium**
- ix) Metallic character of p-block elements depends upon electron population of outer most shell and:  
 \* **Ionization potential** \* Shielding effect  
 \* Electronegativity \* Hydration energy
- x) Which of the following ligand is chelating agent:  
 \* **EDTA** \*  $CN^{-1}$  \*  $NH_3$  \*  $F^{-1}$
- xi) The formula of chromite ore is:  
 \*  $CrO_2Cl_2$  \*  $K_2Cr_2O_7$  \*  **$FeOCr_2O_3$**  \*  $K_2CrO_4$
- xii) The Coordination number of Cobalt in  $[Co(en)_2Cl_2]Cl$  is:  
 \* 3 \* 4 \* **6** \* 8
- xiii) The element which is present in group IV-A and 3<sup>rd</sup> period, its atomic number is:  
 \* 15 \* **14** \* 13 \* 6
- xiv) Oxidation state of oxygen in super oxides is:  
 \* -2 \* +2 \*  **$-1/2$**  \* -1
- xv) 'Li' is powerful reducing agent due to its large value for:  
 \* Ionization potential \* **Hydration enthalpy**  
 \* Electron Affinity \* Electronegativity
- xvi) Aluminium does not react with:  
 \*  $H_2SO_4$  \*  $H_3PO_4$  \*  **$HNO_3$**  \*  $HCl$
- xvii) The ligands which form ring structure with central atom is:  
 \* **Chelate** \* Co-ordination sphere  
 \* Cyclic ligand \* Polydentate ligand
- xviii) Aluminium bronze contains:  
 \* **10% Al and 90% Cu** \* 5% Al and 95% Cu  
 \* 20% Al and 80% Cu \* 90% Al and 10% Cu
- xix) The number of unpaired electrons in ferrous ion is:  
 \* 2 \* 3 \* **4** \* 6

Continued on the next page --2:--.....

# Hsc Exams 1science Sem 2 Old Paper Chemistry

**Jianjun Gao**



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