

## SECTION - I

**Q. 1. Select and write the most appropriate answers from the given alternatives for each sub-question :**

[7]

- (i) In Van Arkel method of refining metal, impure zirconium is converted to unstable volatile compound by heating it with –  
 (a) oxygen (b) chlorine (c) bromine (d) iodine
- (ii) The temperature at which vapour pressure of a liquid becomes equal to the atmospheric pressure is \_\_\_\_\_.  
 (a) melting point (b) boiling point (c) 273 K (d) 373 K
- (iii) Which of the following parameters are correct for triclinic lattice ?  
 (a)  $\alpha = \beta = \gamma = 90^\circ$  and  $a = b = c$  (b)  $\alpha \neq \beta \neq \gamma \neq 90^\circ$  and  $a \neq b \neq c$   
 (c)  $\alpha = \gamma = 90^\circ$ ,  $\beta \neq 90^\circ$  and  $a \neq b \neq c$  (d)  $\alpha \neq \beta \neq \gamma \neq 90^\circ$  and  $a \neq b \neq c$
- (iv) Which mixture is used for respiration by deep sea divers ?  
 (a) He + O<sub>2</sub> (b) Ne + O<sub>2</sub> (c) Ar + O<sub>2</sub> (d) Kr + O<sub>2</sub>
- (v) For the reaction  $2\text{N}_2\text{O}_{5(g)} \longrightarrow 4\text{NO}_{2(g)} + \text{O}_{2(g)}$  in liquid bromine, which of the following rate equation is 'incorrect'  
 (a)  $-\frac{1}{2} \frac{d[\text{N}_2\text{O}_5]}{dt}$  (b)  $-\frac{1}{4} \frac{d[\text{NO}_2]}{dt}$  (c)  $\frac{d[\text{O}_2]}{dt}$  (d)  $\frac{1}{4} \frac{d[\text{NO}_2]}{dt}$
- (vi) For a certain reaction  $\Delta H = -50 \text{ KJ}$  and  $\Delta S = -80 \text{ JK}^{-1}$ , at what temperature does the reaction turn from spontaneous to nonspontaneous.  
 (a) 6.25 K (b) 62.5 K (c) 625 K (d) 6250 K
- (vii) What is the ratio of volumes of H<sub>2</sub> and O<sub>2</sub> liberated during electrolysis of acidified water ?  
 (a) 1 : 2 (b) 2 : 1 (c) 1 : 8 (d) 8 : 1

**Q. 2. Answer any THREE of the following :**

[9]

- (i) Calculate  $\Delta H^\circ$  for the following reactions –



Given that



- (ii) Calculate molarity and molality of 6.3% solution of nitric acid having density  $1.04 \text{ gcm}^{-3}$ . (H = 1, N = 14, O = 16)
- (iii) What is the action of chlorine (Cl<sub>2</sub>) on the following :  
 (a) Cold and dilute caustic soda.  
 (b) Hot and concentrated caustic soda.  
 (c) Potassium bromide solution.
- (iv) Calculate the number of atoms present in 2 gram of crystal which has face-centred cubic (FCC) crystal lattice having edge length of 100 pm and density  $10 \text{ gcm}^{-3}$ .

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