

## SECTION - I

**Q. 1. Select and write the most appropriate answer from the given alternatives for each sub-question :** [7]

- (i) The hybridisation of phosphorus in phosphorus pentachloride is-  
 (a)  $dsp^3$  (b)  $sp^3d$  (c)  $d^2sp^3$  (d)  $sp^3d^2$
- (ii) The rate constant for a first order reaction is  $100\text{ S}^{-1}$ . The time required for completion of 50% of reaction is-  
 (a) 0.0693 milliseconds (b) 0.693 milliseconds  
 (c) 6.93 milliseconds (d) 69.3 milliseconds
- (iii) Silica is added to roasted copper ore during smelting process to remove -  
 (a) ferrous sulphide (b) ferrous oxide (c) cuprous sulphide (d) cuprous oxide
- (iv) 96500 coulombs correspond to the charge on how many electrons ?  
 (a)  $1.6 \times 10^{19}$  (b)  $6.022 \times 10^{20}$  (c)  $6.022 \times 10^{23}$  (d)  $6.022 \times 10^{24}$
- (v) For the reaction :  $\text{Cl}_{2(g)} \longrightarrow 2\text{Cl}_{(g)}$  -  
 (a)  $\Delta H$  is positive,  $\Delta S$  is positive (b)  $\Delta H$  is positive,  $\Delta S$  is negative  
 (c)  $\Delta H$  is negative,  $\Delta S$  is negative (d)  $\Delta H$  is negative,  $\Delta S$  is positive
- (vi) The substance 'X', when dissolved in solvent water gave molar mass corresponding to the molecular formula ' $\text{X}_3$ '. The van't Hoff factor (i) is -  
 (a) 3 (b) 0.33 (c) 1.3 (d) 1
- (vii) The relation  $a \neq b \neq c$  and  $\alpha \neq \beta \neq \gamma \neq 90^\circ$  represents which crystal system ?  
 (a) Orthorhombic (b) Tetragonal (c) Triclinic (d) Monoclinic

**Q. 2. Answer any SIX of the following :**

[12]

- (i) Define ebullioscopic constant. Write its unit.
- (ii) State : (a) Second law of thermodynamics in terms of entropy.  
 (b) Third law of thermodynamics.
- (iii) What is the hybridisation of carbon atom in diamond and graphite ?  
 0.1 mole of Buckminster fullerene contains how many Kg of carbon ?  
 [Atomic mass of carbon = 12]
- (iv) Draw structures of (a) Chlorine trifluoride (b) Chlorine pentafluoride.

# Hsc Chemistry Question Paper 201

**D Kirk**



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## **Table of Contents Hsc Chemistry Question Paper 201**

1. Understanding the eBook Hsc Chemistry Question Paper 201
  - The Rise of Digital Reading Hsc Chemistry Question Paper 201
  - Advantages of eBooks Over Traditional Books
2. Identifying Hsc Chemistry Question Paper 201
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Hsc Chemistry Question Paper 201
  - User-Friendly Interface
4. Exploring eBook Recommendations from Hsc Chemistry Question Paper 201
  - Personalized Recommendations
  - Hsc Chemistry Question Paper 201 User Reviews and Ratings
  - Hsc Chemistry Question Paper 201 and Bestseller Lists

5. Accessing Hsc Chemistry Question Paper 201 Free and Paid eBooks
  - Hsc Chemistry Question Paper 201 Public Domain eBooks
  - Hsc Chemistry Question Paper 201 eBook Subscription Services
  - Hsc Chemistry Question Paper 201 Budget-Friendly Options
6. Navigating Hsc Chemistry Question Paper 201 eBook Formats
  - ePub, PDF, MOBI, and More
  - Hsc Chemistry Question Paper 201 Compatibility with Devices
  - Hsc Chemistry Question Paper 201 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Hsc Chemistry Question Paper 201
  - Highlighting and Note-Taking Hsc Chemistry Question Paper 201
  - Interactive Elements Hsc Chemistry Question Paper 201
8. Staying Engaged with Hsc Chemistry Question Paper 201
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Hsc Chemistry Question Paper 201
9. Balancing eBooks and Physical Books Hsc Chemistry Question Paper 201
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Hsc Chemistry Question Paper 201
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Hsc Chemistry Question Paper 201
  - Setting Reading Goals Hsc Chemistry Question Paper 201
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Hsc Chemistry Question Paper 201
  - Fact-Checking eBook Content of Hsc Chemistry Question Paper 201
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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