

যশোর বোর্ড

- ১০। নিচের কোন বৌপটি পাকা কলায় বিদ্যমান?
 (ক) অ্যামাইল এসিটেট
 (খ) অক্টাইল এসিটেট
 (গ) ইথাইল বিউটাইরেট
 (ঘ) বেনজাইল এসিটেট
- ১১। আলোক সক্রিয় সমানু হওয়ার জন্য অ্যালকেনের সর্বনিম্ন কার্বন সংখ্যা হবে—
 (ক) 7
 (খ) 8
 (গ) 9
 (ঘ) 10
- ১২। স্থির তাপে 0°C তাপমাত্রায় O_2 গ্যাসের আয়তন 3.5L, হলে 20°C তাপমাত্রায় গ্যাসটির আয়তন হবে—
 (ক) 3.25L
 (খ) 3.76L
 (গ) 7.0L
 (ঘ) 8.0L
- ১৩। 0.15M NaOH দ্রবণের ppm ঘনমাত্রা কত?
 (ক) 4000
 (খ) 5000
 (গ) 7000
 (ঘ) 6000
- ১৪। $2\text{FeCl}_3 + \text{SnCl}_2 \rightarrow 2\text{FeCl}_2 + \text{SnCl}_4$ বিক্রিয়াটিতে জারিত হয়েছে কোনটি?
 (ক) FeCl_3
 (খ) FeCl_2
 (গ) SnCl_4
 (ঘ) SnCl_2
- ১৫। কৃষিকাজে সারফেনে জরতিলের বিষুভতার মানদণ্ড কোনটি?
 (ক) TDS
 (খ) COD
 (গ) pH
 (ঘ) BOD
- নিচের উদ্দীপকটি পড় এবং ১৬ ও ১৭ নং প্রশ্নের উত্তর দাও।
 A ধাতুর ক্ষেত্রে $E^{\circ}_{A/A^{2+}} = 0.88\text{V}$
 B ধাতুর ক্ষেত্রে $E^{\circ}_{B/B^{2+}} = -0.35\text{V}$
 $E^{\circ}_{B^{2+}/B} = +0.35\text{V}$
- ১৬। উদ্দীপকের তথ্যসমূহের দ্বারা পঠিত কোষের জন্য—
 i. $E^{\circ}_{\text{cell}} = E^{\circ}_{A/A^{2+}} + E^{\circ}_{B/B^{2+}}$
 ii. $E^{\circ}_{\text{cell}} = E^{\circ}_{A/A^{2+}} + E^{\circ}_{B^{2+}/B}$
 iii. $E^{\circ}_{\text{cell}} = E^{\circ}_{B^{2+}/B} - E^{\circ}_{A^{2+}/A}$
- নিচের কোনটি সঠিক?
 (ক) i ও ii
 (খ) ii ও iii
 (গ) i ও iii
 (ঘ) i, ii ও iii
- ১৭। তড়িৎস্রাবের দ্বারা পঠিত কোষের ক্ষেত্রে প্রযোজ্য হবে—
 (ক) A ধাতু ক্যাথোড হিসেবে ব্যবহৃত হয়
 (খ) A ধাতু B অপেক্ষা অধিক সক্রিয়
 (গ) কোষটির জারমাণ হবে $B/B^{2+} \parallel A^{2+}/A$
 (ঘ) কোষটির বিভব 0.53V

- ১৮। $\text{CH}_3-\text{CH}=\text{CH}-\text{COOH}$ বৌপটির IUPAC নাম হলো—
 (ক) But-1-en-2-oic acid
 (খ) But-2-en-2-oic acid
 (গ) But-2-en-1-oic acid
 (ঘ) Buten-1-oic acid
- ১৯। অম্লসিক এসিডকে NaOH দ্বারা টাইট্রেশনের ক্ষেত্রে ব্যবহৃত নির্দেশকটি হলো—
 (ক) ফেনলফথ্যালিন
 (খ) মিথাইল রেড
 (গ) মিথাইল অরেঞ্জ
 (ঘ) থাইমল ব্লু
- ২০। কোনটি বিজারক পদার্থ?
 (ক) KMnO_4
 (খ) FeSO_4
 (গ) H_2SO_4
 (ঘ) $\text{K}_2\text{Cr}_2\text{O}_7$
- ২১। নিচের কোনটির জন্য $\frac{d}{dt}$ প্রযোজ্য?
 (ক) মোলারিটি
 (খ) মোলপরিমাণ
 (গ) নরমালিটি
 (ঘ) ফরমালিটি
- ২২। তড়িৎ বিশ্লেষণকালে কোনটি অণু চার্জযুক্ত হবে?
 (ক) H^+
 (খ) Pb^{2+}
 (গ) Cu^{2+}
 (ঘ) Na^+
- ২৩। $\text{RCH}_2\text{CH}_2\text{X} \xrightarrow{\text{KCN}} \text{X} \xrightarrow[\text{H}^+]{\text{H}_2\text{O}} \text{Y}$
 Y হলো—
 (ক) ক্রিটোল
 (খ) অ্যালকোহল
 (গ) অ্যালডিহাইড
 (ঘ) কার্বোক্সিলিক এসিড
- ২৪। $\text{CH}_3\text{CH}_2\text{Cl} \xrightarrow[\text{তড়িৎ বিদ্যুৎ}]{\text{Na}} \text{B}$
 $\text{B} \xrightarrow{\text{KOH(aq)}} \text{C}$
- B ও C উৎপাদনকারের সংকেত হল—
 (ক) $\text{CH}_3(\text{CH}_2)_2\text{CH}_3$ ও $\text{CH}_3\text{CH}_2\text{OH}$
 (খ) $\text{CH}_3\text{CH}_2\text{CH}_3$ ও $\text{CH}_2=\text{CH}_2$
 (গ) $\text{CH}_3(\text{CH}_2)_2\text{CH}_3$ ও $\text{CH}_2=\text{CH}_2$
 (ঘ) $\text{CH}_3\text{CH}_2\text{CH}_3$ ও $\text{CH}_3\text{CH}_2\text{OH}$
- ২৫। কোনটি অগ্রতিলম অ্যালকিন?
 (ক) $\text{CH}_2=\text{CH}_2$
 (খ) $\text{CH}_3-\text{CH}=\text{CH}-\text{CH}_3$
 (গ) $\text{CH}_2=\text{CH}-\text{CH}_3$
 (ঘ) $\text{ClCH}=\text{CHCl}$

Hsc 2chamestie 2nd Paper Question

JA Banks



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