

On your own

1. One gallon of gasoline can be considered to contain about 25 moles of octane (C_8H_{18}). Octane burns according to the following equation:



- (a) How many moles of oxygen gas must be used to burn a gallon of gasoline? $312.5 \text{ mol } O_2$
 (b) How many moles of carbon dioxide are formed? $200 \text{ mol } CO_2$
 (c) How many grams of carbon dioxide are formed? $8500 \text{ g } CO_2$

2. Many statues are made of marble (calcium carbonate). Acid rain contains sulfuric acid, which reacts with marble according to the following equation:



- (a) How many moles of marble will be consumed if 5 mol H_2SO_4 fall onto the statue? $5 \text{ mol } CaCO_3$
~~How many moles of CO_2 are produced when 12 g marble are reacted at STP?~~
 (c) How many grams of the statue will be consumed if 500 g H_2SO_4 fall onto the statue? $510 \text{ g } CaCO_3$

3. Plants convert carbon dioxide, water, and energy from the sun into sugar ($C_6H_{12}O_6$) and dioxygen in the process of photosynthesis.



- (a) How many grams of sugar are produced from 1000g (one liter) of water? $1667 \text{ g } C_6H_{12}O_6$
~~How many liters of CO_2 are needed to react with one liter of water?~~

4. Thermite reactions have been used in the past for welding railroad rails, making incendiary bombs, and igniting solid-fuel rocket motors.



- (a) How many grams of iron (III) oxide will be needed to react in order to form 15.0 g iron? $21.4 \text{ g } Fe$
 (b) How many grams of aluminum will be needed to react in order to form 15.0 g iron? $7.2 \text{ g } Al$
 (c) How many grams of aluminum oxide will be formed if 1.63×10^{23} atoms iron are formed? $13.8 \text{ g } Al_2O_3$

5. Calcium cyanamide ($CaCN_2$) is used as a fertilizer. The water in soil react with the fertilizer to deposit NH_3 and $CaCO_3$ into the ground.



- (a) How many grams of NH_3 are deposited into the soil if 250 kg $CaCN_2$ are used? $106250 \text{ g } NH_3$
 (b) How many grams of water from the soil must react with 250 kg $CaCN_2$? $118750 \text{ g } H_2O$

Hs Chemistry Pogil Activity Basic Stoichiometry On Your Own

Tao Wei



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