

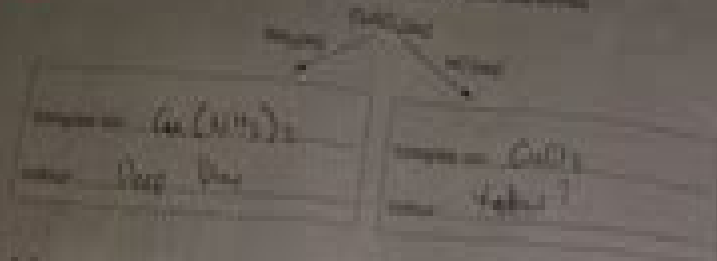
Question 1

Answer all the questions

The reaction is about the synthesis of copper complexes and ligands.

(a) The reaction shows the synthesis of a copper complex.

In the boxes, write the formula and name of the ligands and metal.



(b) Cu^{2+} has two d-orbitals in d_{sp^2} hybridization state and two water molecules. The d_{sp^2} hybridization is a tetrahedral ligand.

The resulting formula of the complex is shown in Fig. 1.1 below.



Fig. 1.1

(c) What is meant by the term tetrahedral ligand?

A ligand that forms four bonds with central metal ion.

(d) The reaction is shown in Fig. 1.2 below.

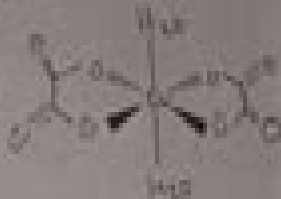

The reaction is about the synthesis of a copper complex. The charge has been given.



Complex A

Complete the 3D diagram of the other two geometries of A. You do not need to include any charges.

Indicate with ticks whether the stereoisomers are cis, trans, optical or a combination of these types.

<p>Stereoisomer</p>														
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(e) What is the empirical formula, including the charge, of the complex in A?

$[Cu(NH_3)_4]^{2+}$

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Jicheng Xie



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Table of Contents F32may 2013 Ocr Chemistry Unofficial Paper

1. Understanding the eBook F32may 2013 Ocr Chemistry Unofficial Paper
 - The Rise of Digital Reading F32may 2013 Ocr Chemistry Unofficial Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying F32may 2013 Ocr Chemistry Unofficial Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an F32may 2013 Ocr Chemistry Unofficial Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from F32may 2013 Ocr Chemistry Unofficial Paper
 - Personalized Recommendations
 - F32may 2013 Ocr Chemistry Unofficial Paper User Reviews and Ratings
 - F32may 2013 Ocr Chemistry Unofficial Paper and Bestseller Lists
5. Accessing F32may 2013 Ocr Chemistry Unofficial Paper Free and Paid eBooks
 - F32may 2013 Ocr Chemistry Unofficial Paper Public Domain eBooks
 - F32may 2013 Ocr Chemistry Unofficial Paper eBook Subscription Services

- F32may 2013 Ocr Chemistry Unofficial Paper Budget-Friendly Options
- 6. Navigating F32may 2013 Ocr Chemistry Unofficial Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - F32may 2013 Ocr Chemistry Unofficial Paper Compatibility with Devices
 - F32may 2013 Ocr Chemistry Unofficial Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of F32may 2013 Ocr Chemistry Unofficial Paper
 - Highlighting and Note-Taking F32may 2013 Ocr Chemistry Unofficial Paper
 - Interactive Elements F32may 2013 Ocr Chemistry Unofficial Paper
- 8. Staying Engaged with F32may 2013 Ocr Chemistry Unofficial Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers F32may 2013 Ocr Chemistry Unofficial Paper
- 9. Balancing eBooks and Physical Books F32may 2013 Ocr Chemistry Unofficial Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection F32may 2013 Ocr Chemistry Unofficial Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine F32may 2013 Ocr Chemistry Unofficial Paper
 - Setting Reading Goals F32may 2013 Ocr Chemistry Unofficial Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of F32may 2013 Ocr Chemistry Unofficial Paper
 - Fact-Checking eBook Content of F32may 2013 Ocr Chemistry Unofficial Paper
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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