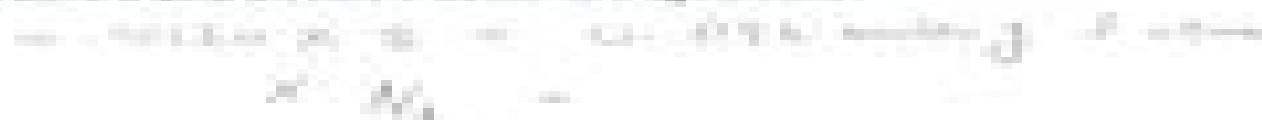
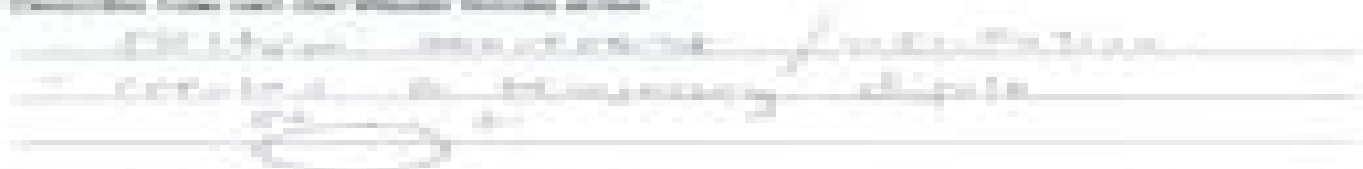


100) Carbonic acid reacts as a buffer at  $H_2CO_3$  molecules. What  $H_2CO_3$  molecule is a ring of single atoms. How many atoms of sulfur are there in a molecule of  $H_2CO_3$  molecule?



101) The only intermediate forms in acid buffer are not the (H<sub>2</sub>CO<sub>3</sub>).

102) Describe how the (H<sub>2</sub>CO<sub>3</sub>) forms acid.



103) Suggest why there are no other intermediate forms in acid buffer.



104) Carbonic acid is a compound of sulfur (sulfur) having (sulfur) atoms.

105) Describe how the (H<sub>2</sub>CO<sub>3</sub>) forms acid.

106) What is the composition number of sulfur in  $H_2CO_3$  molecule?



# F321 Chemistry May 2013

**Mike Smith**



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