

F321: Atoms, Bonds and Groups

Structure & Bonding – Mark Scheme

1. (i) (Electrostatic) **attraction** between oppositely charged **ions** ✓

IGNORE force

IGNORE references to transfer of electrons

MUST be ions, not particles

1

- (ii) Mg shown with either 8 or 0 electrons

AND

S shown with 8 electrons **with** 2 crosses and 6 dots (or vice versa) ✓

Correct charges on both ions ✓



Mark charges on ions and electrons independently

For first mark, if 8 electrons are shown around the Mg then 'extra electrons' around S must match the symbol chosen for electrons around Mg

Shell circles not required

IGNORE inner shell electrons

Brackets are not required

2

[2]

2. (i) Electron pairs in covalent bonds shown correctly using dots and crosses in a molecule of the F_2O ✓

Lone pairs correct on O and both F atoms ✓



Must be 'dot-and-cross'

circles for outer shells *NOT* needed

IGNORE inner shells

Non-bonding electrons of O do not need to be shown as pairs

Non-bonding electrons of F do not need to be shown as pairs

2

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C Cleary



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