$pH = -log_{10}[H^*]$ 

 $\frac{1}{2} F_{\alpha}(g) + g^{\alpha}$ 

# Chemistry

#### DATA SHEET

Avogadro constant, NA		6.022	$\times 10^{23} \text{ mol}^{-1}$
Volume of I mole ideal gas: at			
	at 0°C (273.15 K)	22.71	L.
	at 25°C (298.15 K)	24.79	L
	25°C (298.15 K), K <sub>w</sub>		
Specific heat capacity of water		$4.18 \times$	c 10 <sup>3</sup> J kg <sup>-1</sup> K <sup>-1</sup>

#### Some useful formulae

Some standard potentials

 $\Delta H = -m C \Delta T$ 

2.89 W

AND DESCRIPTION OF		ar o poetarears	
K* + e*	apple.	BCGr0	-2.94 W
Ba2+ + 2e-	ng-Mr.	Bo(x)	-2.91 W
Ca2+ + 2e-	mark.	Ca(x)	-2.87 W
Na* + e*	marks.	Nia(a)	-2.71 V
Mg2+ + 2e"	ngoliti .	Mg(x)	-2.36 V
All * 3e*	spills.	All(x)	-1.68 V
Mas2+ 2e	web.	Mim(x)	-1.18 V
H <sub>2</sub> O + e <sup>-</sup>	age to	$\frac{1}{2}H_2(g) + OH^-$	-0.83 V
$2n^{2s} + 2e^{-}$	100	Zm(x)	-0.76 V
Fe <sup>3+</sup> + 2e <sup>-</sup>	appeter.	PinCx9	-0.44 W
Nii <sup>3+</sup> = 2e <sup>-</sup>	1975	Nikoro	-0.24 V
Sm2+ + 2e-	Spide.	Sin(x)	-0.14 W
Pb <sup>2+</sup> + 2e <sup></sup>	spile.	P%(x):	-0.13 V
16" + e"	apple 1	$\frac{1}{2}10_{2}(g)$	0.00 V
SO <sub>4</sub> 2- + 4H* + 2e*	neth .	$SO_2(seq) = 2H_2O$	0.16 V
Cu2= + 2e-	reprint.	Checks	0.34 W
$\frac{1}{2}O_2(g) + H_2O + 2e^{-g}$	neste.	2OH-	0.40 V
Cu* + e*	1997	Cocce	0.52 V
$\frac{1}{2}I_{\alpha}(x) = \alpha^{-}$	nests.	II-	0.54 W
±1,(aq) + e	neth-	125	0.62 V
Fe3 e	spots	Fe <sup>3</sup> *	0.77 V
Ag" - e"	marks.	Ag(r)	0.80 V
$\frac{1}{2}Be_2(0) = e^{-}$	- spile	Br.	1.08 V
	1990	Br.	1:30 W
$\frac{1}{2}O_2(g) + 2H^* + 2e^-$	aprile .	H <sub>2</sub> O	1.23 W
\$Cl <sub>2</sub> (g) + e <sup>-</sup>	400	CIT	1.36 V
1 Ce2O22- + 7H* + 3e-	april 1	$Cr^{3-} = \frac{7}{2}H_2O$	1.36 W
Cl <sub>2</sub> (mq) + e	spile.	CIT	3.40 W
MmO <sub>4</sub> " + 8H" + 5e"	spik.	$Mn^{2+} + 4H_dO$	1.51 W

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