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C

Where is the hydrophilic (attracted to water) region of the molecule?

- (A) Region 1
- (B) Region 2
- (C) Region 3
- (D) The three regions are equally hydrophilic



C

Which molecule is most soluble in water?



D.

A solution of NaCl in water has a concentration of 20.5% by mass. What is the molal concentration of the solution?

Molar Mass NaCl = 58.44 g/mol

- (A) 0.205 m
- (B) 0.258 m
- (C) 3.51 m
- (D) 4,41 m

8

What is the mole fraction of water in 200 g of 89% (by mass) ethanol, C2H5OH?

Molar Mass C2H3OH = 46 g/mol

(A) 0.11

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Keith S Taber

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Preparing for Chemistry Teaching Festo Kayima,2025-08-13 This textbook is a comprehensive chemistry didactics resource for chemistry teacher educators chemistry teachers and trainees It provides research grounded and practical based pedagogical experiences examples and frameworks for chemistry teachers as well as a foundation for planning and implementing productive chemistry lessons The book provides a conceptual and practical roadmap illuminating which didactic knowledge elements are relevant for becoming a chemistry teacher The book starts off with a pedagogically laden however experience based justification for the relevance of chemistry didactics and then progressively breaks down the different knowledge elements that form a complete set of the didactic knowledge and skill elements a teacher needs for teaching Concrete examples are provided to allow the reader to operationalize the ideas and concepts presented in the book The structure of the chapters enables the reader to engage progressively and actively with its contents and provided examples allowing a deep understanding of the diverse links between the presented topics forming a complete set of the didactic knowledge and skills relevant for successful chemistry teaching

Tests in Print Oscar Krisen Buros, 2006

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