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Evolving Nature of Objectivity in the History of Science and its Implications for Science Education Mansoor Niaz, 2017-10-26 This book explores the evolving nature of objectivity in the history of science and its implications for science education It is generally considered that objectivity certainty truth universality the scientific method and the accumulation of experimental data characterize both science and science education Such universal values associated with science may be challenged while studying controversies in their original historical context The scientific enterprise is not characterized by objectivity or the scientific method but rather controversies alternative interpretations of data ambiguity and uncertainty Although objectivity is not synonymous with truth or certainty it has eclipsed other epistemic virtues and to be objective is often used as a synonym for scientific Recent scholarship in history and philosophy of science has shown that it is not the experimental data Baconian orgy of quantification but rather the diversity plurality in a scientific discipline that contributes toward understanding objectivity History of science shows that objectivity and subjectivity can be considered as the two poles of a continuum and this dualism leads to a conflict in understanding the evolving nature of objectivity The history of objectivity is nothing less than the history of science itself and the evolving and varying forms of objectivity does not mean that one replaced the other in a sequence but rather each form supplements the others This book is remarkable for its insistence that the philosophy of science and in particular that discipline s analysis of objectivity as the supposed hallmark of the scientific method is of direct value to teachers of science Meticulously yet in a most readable way Mansoor Niaz looks at the way objectivity has been dealt with over the years in influential educational journals and in textbooks it s fascinating how certain perspectives fade while basic questions show no sign of going away There are few books that take both philosophy and

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Reconstruction of Wave-Particle Duality and its Implications for General Chemistry Textbooks Mansoor

Niaz,Cecilia Marcano,2012-04-26 It goes without saying that atomic structure including its dual wave particle nature cannot be demonstrated in the classroom Thus for most science teachers especially those in physics and chemistry the textbook is their key resource and their students core source of information Science education historiography recognizes the role played by the history and philosophy of science in developing the content of our textbooks and with this in mind the authors analyze more than 120 general chemistry textbooks published in the USA based on criteria derived from a historical reconstruction of wave particle duality They come to some revealing conclusions including the fact that very few textbooks discussed issues such as the suggestion by both Einstein and de Broglie and before conclusive experimental evidence was available that wave particle duality existed Other large scale omissions included de Broglie s prescription for observing this duality and the importance of the Davisson Germer experiments as well as the struggle to interpret the experimental data they were collecting Also untouched was the background to the role played by Schr dinger in developing de Broglie s ideas The authors argue that rectifying these deficiencies will arouse students curiosity by giving them the opportunity to engage creatively with the content of science curricula They also assert that it isn t just the experimental data in science that matters but the theoretical insights and unwonted inspirations too In addition the controversies and discrepancies in the theoretical and experimental record are key drivers in understanding the development of science as we know it today *Chemistry for*

Cooks Sandra C. Greer,2023-01-10 A fun approach to teaching science that uses cooking to demonstrate principles of chemistry for undergraduate students who are not science majors high school students culinary students and home cooks How does an armload of groceries turn into a culinary masterpiece In this highly accessible and informative text Sandra C Greer takes students into the kitchen to show how chemistry with a dash of biology and physics explains what happens when we cook *Chemistry for Cooks* provides all the background material necessary for nonscientists to understand essential chemical processes and to see cooking as an enjoyable application of science Greer uses a variety of practical examples including recipes to instruct readers on the molecular structure of food the chemical reactions used in cooking to change the nature of food and the essentials of nutrition and taste She also offers kitchen hints and exercises based on the material in each chapter plus do it yourself projects to encourage exploration of the chemistry that takes place when we cook food Features Perfect for science courses aimed at non science majors does not require prior knowledge of chemistry physics or biology Equally useful for general readers home and professional cooks and culinary students Topics include what matter is made of how the structure of matter is altered by heat how we treat food in order to change its microscopic structure why particular procedures or methods are used in the kitchen and how to think critically about various cooking methods A reference section at the end of each chapter points readers to resources for further study Additional online resources include

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Human Chemistry (Volume Two) Libb Thims, 2007-09-01 Volume two begins with Goethe's theories of affinities i.e. the chemical reaction view of human life in 1809 This is followed by the history of how the thermodynamic 1876 and quantum 1905 revolutions modernized chemistry such that affinity the force of reaction is now viewed as a function of thermodynamic free energy reaction spontaneity and quantum valency bond stabilities The composition energetic state dynamics and evolution of the human chemical bond A B is the centerpiece of this process The human bond is what gives yields and takes absorbs energy in life The coupling of this bond energy driven by periodic inputs of solar photons thus triggering activation energies and entropies connected to the dynamical work of life is what quantifies the human reaction process This is followed by topics including mental crystallization template theory LGBT chemistry chemical potential Le Chatelier's principle Muller dispersion forces and human thermodynamics

Principles of Qualitative Inorganic Analysis: Precipitation, Separation and Identification of Cations Huda S. Alhasan, Nadiyah Alahmadi, 2021-01-26 This book provides notes for basic laboratory experiments in qualitative analysis of cations The book introduces readers to basic methods and laboratory safety Subsequent chapters cover six groups of cations Each chapter explains important details that are required to understand how a particular analytical method works for detecting cations in samples starting from sedimentation and ending with the identification Key Features Simple reader friendly format introductory notes and summary Covers several groups of metals Appendix for handy reference with tables and references This is a useful textbook for early chemistry students and teachers as it equips the readers with sufficient information required to analyze chemical samples and deduce the presence of specific cations as part

of laboratory coursework **General Chemistry** Darrell Ebbing, Steven D. Gammon, 2007-12-03 The Ninth Edition of this market leading text has been substantially revised to meet the rapidly changing instructional demands of General Chemistry professors Known for its carefully developed thoroughly integrated step by step approach to problem solving General Chemistry helps students master quantitative skills and build a lasting conceptual understanding of key chemical concepts The Ninth Edition retains this hallmark approach and builds upon the conceptual focus through key new features and revisions Important Notice Media content referenced within the product description or the product text may not be available in the ebook version **Physico-chemical Aspects of Textile Coloration** Stephen M. Burkinshaw, 2016-02-08 The

production of textile materials comprises a very large and complex global industry that utilises a diverse range of fibre types and creates a variety of textile products As the great majority of such products are coloured predominantly using aqueous dyeing processes the coloration of textiles is a large scale global business in which complex procedures are used to apply different types of dye to the various types of textile material The development of such dyeing processes is the result of substantial research activity undertaken over many decades into the physico chemical aspects of dye adsorption and the establishment of dyeing theory which seeks to describe the mechanism by which dyes interact with textile fibres Physico Chemical Aspects of Textile Coloration provides a comprehensive treatment of the physical chemistry involved in the dyeing of the major types of natural man made and synthetic fibres with the principal types of dye The book covers fundamental aspects of the physical and chemical structure of both fibres and dyes together with the structure and properties of water in relation to dyeing dyeing as an area of study as well as the terminology employed in dyeing technology and science contemporary views of intermolecular forces and the nature of the interactions that can occur between dyes and fibres at a molecular level fundamental principles involved in dyeing theory as represented by the thermodynamics and kinetics of dye sorption detailed accounts of the mechanism of dyeing that applies to cotton and other cellulosic fibres polyester polyamide wool polyacrylonitrile and silk fibres non aqueous dyeing as represented by the use of air organic solvents and supercritical CO₂ fluid as alternatives to water as application medium The up to date text is supported by a large number of tables figures and illustrations as well as footnotes and widespread use of references to published work The book is essential reading for students teachers researchers and professionals involved in textile coloration **Waking the Power Within**

Thermodynamics and the Human Battery Professor Ibrahim Elnoshokaty , The sci fi film The Matrix introduces a fascinating premise where humans function as energy sources for an advanced machine society In this fictional world human bodies are maintained in a state of suspended animation while their minds exist in a virtual reality allowing machines to extract their bioelectric thermal and kinetic energy This article investigates the scientific feasibility of utilizing humans as a power source by applying thermodynamic principles According to the first law of thermodynamics the energy required to sustain human life would result in a net energy loss for the machines The second law indicates that the system's entropy

would rise rendering it an inefficient energy strategy Furthermore the energy output of a human body even if fully utilized would be inadequate to meet the machines energy demands More efficient alternatives for the machines would include other biological power sources and energy harvesting techniques such as solar or nuclear power The article concludes that while the concept of human batteries serves as an engaging storytelling element it is not a scientifically viable solution for the machines energy requirements The machines choice to preserve human life may be motivated by other factors such as leveraging their collective cognitive abilities for computational purposes or adhering to an ethical code that prohibits the complete annihilation of humanity This investigation aims to fill the gap by providing a detailed thermodynamic analysis of the energy expenditure required to sustain human life in a suspended animation state and the inefficiency of this system as an energy source for machines a facet previously unexplored By elucidating the thermodynamic constraints of human based energy sources this study not only challenges a popular sci fi narrative but also enriches our understanding of bioenergetic processes and their implications for future energy harvesting technologies

The Chemistry Connection: From Atoms to Applications Dr. Sarika Arora, 2024-09-16 Whether you're an avid student or an inquisitive learner The Chemistry Connection From Atoms to Applications is your key to unlocking the amazing world of chemistry This book breaks down the basic components of matter atoms molecules and chemical reactions into clear explanations simplifying complicated ideas This book makes the connections demonstrating how chemistry affects everything around us from the smallest particles to the most significant applications in daily life You will learn about the amazing mechanisms that underpin everything in our world including the food we consume the technologies we use and even the surrounding natural beauty Through lucid illustrations meaningful comparisons and useful advice The Chemistry Connection makes science approachable and interesting for all readers This book provides a thorough exploration of the fundamentals of chemistry and its practical applications making it ideal for anybody wishing to brush up on their knowledge develop a better understanding of the topic or just quench their curiosity Explore and learn how atoms relate to your surroundings

Calculations in Chemical Kinetics for Undergraduates Eli Usheunepa Yunana, 2022-06-14 Calculations in Chemical Kinetics for Undergraduates aims to restore passion for problem solving and applied quantitative skills in undergraduate chemistry students Avoiding complicated chemistry jargon and providing hints and step wise explanations in every calculation problem students are able to overcome their fear of handling mathematically applied problems in physical chemistry This solid foundation in their early studies will enable them to connect fundamental theoretical chemistry to real experimental applications as graduates Additional Features Include Contains quantitative problems from popular physical chemistry references Provides step by step explanations are given in every calculation problem Offers hints to certain problems as points to note to enable student comprehension Includes solutions for all questions and exercises This book is a great resource for undergraduate chemistry students however the contents are rich and useful to even the graduate chemist that has passion for applied problems in physical

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