

CASCADED MC QUESTIONS: GENERAL CHEMISTRY II

PETRA JOHNS HIGHER SECONDARY SCHOOL

1. Which among the intermolecular forces is the weakest?
 - A. Hydrogen Bond
 - B. Ion-Dipole Forces
 - C. Dipole-Dipole Forces
 - D. Dispersion Forces
2. In which of the following would dipole-dipole forces exist?
 - A. Butane
 - B. Methane
 - C. CH_2Cl_2
 - D. CH_3OH
3. The vapour pressure of alcohol at 40 °C is 120 mmHg. What would be the vapour pressure at 30 °C?
 - A. 120 mmHg
 - B. Greater than 120 mmHg
 - C. Less than 120 mmHg
 - D. Cannot be determined
4. Which of the following statements about the following process will be correct when it occurs?
 - A. It is endothermic
 - B. It is exothermic
 - C. It should take some time to reach equilibrium
 - D. It depends on the nature of the gas
5. Which of the following would have the smallest ΔH_{vap} ?
 - A. H_2O
 - B. CH_3OH
 - C. H_2
 - D. C_2H_6
6. Which of the following reactions would be the most exothermic?
 - A. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$
 - B. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}_2$
 - C. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$
 - D. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}_2$
7. Which of the following describes Henry's Law?
 - A. Partial pressure increases with the solubility of gases.
 - B. Partial pressure increases with the solubility of gases.
 - C. Partial pressure increases with the solubility of solids.
 - D. Partial pressure increases with the solubility of solids.
8. How much of gas is dissolved in 1.00 L of water at 25 °C?
 - A. 1.00 mol
 - B. 1.00 g
 - C. 1.00 mol/L
 - D. 1.00 mol/L
9. Water's vapour pressure at 20 °C is 17.5 mmHg. The temperature was held constant while adding NaCl , resulting in a new vapour pressure of 12.0 mmHg. What is the new temperature?
 - A. 10.0 °C
 - B. 15.0 °C
 - C. 20.0 °C
 - D. 25.0 °C
10. Which of the following is not a colligative property?
 - A. Boiling point
 - B. Freezing point
 - C. Osmotic pressure
 - D. Density

General Chemistry Diagnostic Test

Hussin A.Rothana



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,2019-02-18 Science Inquiry Argument and Language describes research that has focused on addressing the issue of embedding language practices within science inquiry through the use of the Science Writing Heuristic approach In recent years much attention has been given to two areas of science education scientific argumentation and science literacy The research into scientific argument have adopted different orientations with some focusing on science argument as separate to normal teaching practices that is teaching students about science argument prior to using it in the classroom context while others have focused on embedding science argument as a critical component of the inquiry process The current emphasis on science literacy has emerged because of greater understanding of the role of language in doing and reporting on science Science is not viewed as being separate from language and thus there is emerging research emphasis on how best to improving science teaching and learning through a language perspective Again the research orientations are parallel to the research on scientific argumentation in that the focus is generally between instruction separate to practice as opposed to embedding language practices within the science classroom context *Teaching Science Online* Dietmar

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