

CHAPTER 14: COVALENT BONDING: ORBITALS

PROBLEMS WITH LEWIS STRUCTURE

- Provides no indication of the **shapes** of the molecules
- O₂ would be predicted to be diamagnetic, but it is **paramagnetic**

Paramagnetic = has unpaired electrons.



The Lewis structure of O₂ has an O=O double bond with **no unpaired electrons**, i.e. diamagnetic.

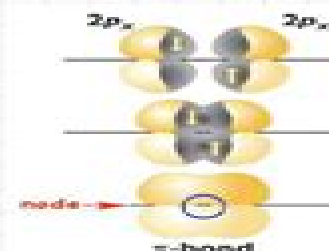
- Many compounds cannot be represented by a single structure
- Resonance structures **differ only in position of electrons** not the nuclei.
- Result in notions of hypervalent compounds and "expansion of octet"
- Can't explain molecules such as B₂H₆ (electron deficient molecules)

VALENCE-BOND THEORY "LOCALIZED ELECTRON MODEL"

- Covalent bond is created by the overlap of orbitals on two atoms

TYPES OF OVERLAP: σ and π Bonds

- Bonding in H₂**
 - Sausage shaped orbital, bond called **σ -orbital**
 - Called σ orbital because it has **no node** and is **cylindrically symmetrical**
 - Electron density is **concentrated between two nuclei**, but is still highest at the nuclei themselves.
- Bonding in N₂
 - σ orbital** is obtained by overlap of **two p_x-orbitals**
- Perpendicular p orbitals overlap in a side-by-side manner
- The resulting orbital has two lobes of electron density and is called a **π -bond**
- Called π -orbital because it **has a node** and, when viewed down the internuclear axis, it looks like a p-orbital
- π -bond has **no electron density between the nuclei along the bond axis**



General Chemistry Chapter 14 Covalent Bonding

Orbitals Practice

Judd E. Hollander



General Chemistry Chapter 14 Covalent Bonding Orbitals Practice:

Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples, and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step by step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid base reactions, which often persists through working with acid base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshmen in STEM fields. *MCAT General Chemistry Review 2023-2024* Kaplan Test Prep, 2022-07-05 Kaplan's *MCAT General Chemistry Review 2023-2024* offers an expert study plan, detailed subject review, and hundreds of online and in-book practice questions, all authored by the experts behind the MCAT prep course that has helped more people get into medical school than all other major courses combined. Prepping for the MCAT is a true challenge; Kaplan can be your partner along the way, offering guidance on where to focus your efforts and how to organize your review. This book has been updated to match the AAMC's guidelines precisely, so no more worrying about whether your MCAT review is comprehensive. The Most Practice: More than 350 questions in the book and access to even more online, more practice than any other MCAT general chemistry book on the market. The Best Practice: Comprehensive general chemistry subject review is written by top-rated, award-winning Kaplan instructors. Full-color 3-D illustrations from *Scientific American*, charts, graphs, and diagrams help turn even the most complex science into easy-to-visualize concepts. All material is vetted by editors with advanced science degrees and by a medical doctor. Online resources, including a full-length practice test, help you practice in the same computer-based format you'll see on Test Day. Expert Guidance: High-yield badges throughout the book identify the topics most frequently tested by the AAMC. We know the test: The Kaplan MCAT team has spent years studying every MCAT-related document available. Kaplan's expert psychometricians ensure our practice questions and study materials are true to the test. **MCAT General Chemistry Review 2022-2023** Kaplan Test Prep, 2021-07-06 Kaplan's *MCAT General Chemistry Review 2022-2023* offers an expert study plan, detailed subject review, and hundreds of online and in-book practice questions, all authored by the experts behind the MCAT prep course that has helped more people get into medical school than all other major courses.

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