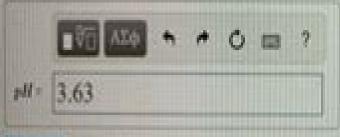


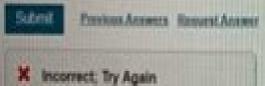
Acid Base Homework

Exercise 16.69

## · Part A

If 20.0 mL of glacial acetic acid (pure  $HC_1H_2O_2$ ) is distend to 1.50 L with water, what is the pH of the resulting solution? The density of glacial acetic acid is 1.05 g/mL. Express your answer to two decimal places.







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Thirumalaisamy P. Velavancorresponding

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A Course in General Chemistry William Crowell Bray, Wendell Mitchell Latimer, 1923

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