

Chem 100 Practice Test: Thermodynamics 3.5 – 5.5, 11.5**Equations**

$$\Delta G = \Delta H - T\Delta S$$

$$\Delta G = \Delta H_{\text{sys}} + P\Delta V_{\text{sys}}$$

$$\Delta G = \Delta H_{\text{sys}}$$

$$\Delta H_{\text{sys}} = \sum \text{products} - \sum \text{reactants}$$

$$\Delta S = \Delta S_{\text{sys}} + \Delta S_{\text{surr}}$$

Properties of water

$$\Delta H_{\text{vap}} = 40.7 \text{ kJ/mol}$$

$$\Delta H_{\text{fus}} = 6.01 \text{ kJ/mol}$$

$$\Delta H_{\text{cond}} = -40.7 \text{ kJ/mol}$$

$$\Delta H_{\text{sub}} = 46.7 \text{ kJ/mol}$$

$$\Delta H_{\text{liq}} = 44.0 \text{ kJ/mol}$$

Multiple choice

Multiple choice and select-all-that-apply questions follow. (The selections will be noted)

(A) spontaneous

(B) system

(C) path

(D) enthalpy

(E) heat capacity

(F) specific heat

(G) heating curve

(H) heat of vaporization

(I) heat of fusion

- ☒ 1. spontaneity of thermodynamic favor the temperature of ΔG will shift by $T\Delta S$.
- ☒ 2. full work of energy
- ☒ 3. spontaneity of thermodynamic change the temperature of ΔG will shift by $T\Delta S$.
- ☒ 4. thermodynamic favorability that thermodynamic favorability during a chemical reaction process
- ☒ 5. balance a system at constant pressure
- ☒ 6. the enthalpy change caused by structure modification
- ☒ 7. the energy relationship and control at the reaction point

Multiple choice (continued)

8. thermodynamic favorability is a path

☐ A. thermodynamic favorability

☒ B. thermodynamic favorability

☐ C. thermodynamic favorability

☐ D. thermodynamic favorability

9. thermodynamic favorability is a thermodynamic favorability

☐ A. thermodynamic favorability

☐ B. thermodynamic favorability

☐ C. thermodynamic favorability

☒ D. thermodynamic favorability

10. thermodynamic favorability is a thermodynamic favorability

☒ A. thermodynamic favorability

☐ B. thermodynamic favorability

☐ C. thermodynamic favorability

☐ D. thermodynamic favorability

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Chao Zhang



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