

09/08/2020

# ## Chapter 1: General Chem Review - Part 1 ##

- 1.1 Reactions: a) molecules collide  
b) bonds broken/formed

Organic compounds → compounds w carbon atoms

- 1.2 Structural Theory: substances are defined by a specific arrangement  
compounds differ based on the way the atoms are bonded

Constitutional isomer: same molecular formula but different structures

Atoms commonly bonded to C: N, O, H & halides

tetravalent  
(4 bonds)

trivalent  
(3 bonds)

divalent  
(2 bonds)

monovalent  
(one bond)

- 1.3 Covalent bond: pair of  $e^-$  shared by two atoms

Energy released when forming bond → equilibrium internuclear distance

internuclear repulsion if contracted further  
(↑ energy)

(bond length)

internuclear distance

energy @ equilibrium  
= bond strength

↑ ↑ space = ↓ energy of electrons

Potential Energy & stability: ↓ PE = most stable

What keeps bond @ optimal length?

+ nuclei & - electron → attractive forces

+ nuclei & + nuclei → repulsive forces ← - electron & - electron

Lewis Structure Examples:



- 1.4 Formal charge: valence electrons it has vs. valence electrons need to be neutral  
eg.  $\text{CH}_3\text{O}$



C has 4 valence electrons → no formal charge  
Needs 4 more to be neutral

O has 7 electrons → 1- formal charge  
Needs 1 less to be neutral

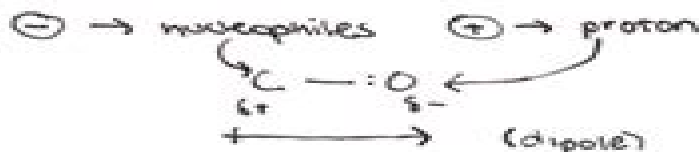
- 1.5 Electronegativity: increase L to R & B to T  
F is most EN

↑ carbon usually forms polar covalent bonds

Covalent bond: EN difference < 0.5

Polar covalent bond: EN difference 0.5 - 1.7

Ionic bond: EN difference > 1.7



# Gen Chem 09 10 Sem 1 Review Answers

**Siddappa N.Byrareddy**



## Gen Chem 09 10 Sem 1 Review Answers:

*Survival Guide to Organic Chemistry* Patrick E. McMahon, Bohdan B. Khomtchouk, Claes Wahlestedt, 2016-12-19 The Survival Guide to Organic Chemistry Bridging the Gap from General Chemistry enables organic chemistry students to bridge the gap between general chemistry and organic chemistry It makes sense of the myriad of in depth concepts of organic chemistry without overwhelming them in the necessary detail often given in a complete organic chemistry text Here the topics covered span the entire standard organic chemistry curriculum The authors describe subjects which require further explanation offer alternate viewpoints for understanding and provide hands on practical problems and solutions to help master the material This text ultimately allows students to apply key ideas from their general chemistry curriculum to key concepts in organic chemistry Key Features Reviews key general chemistry concepts and techniques adapted for application to important organic principles Provides practical guidance to help students make the notoriously well known and arduous transition from general chemistry to organic chemistry Explains organic concepts and reaction mechanisms generally expanding the focus on how to understand each step from a more intuitive viewpoint Covers concepts that need further explanation as well as those that summarize and emphasize key ideas or skills necessary in this field An added bonus is help with organizing principles to make sense of a wide range of similar reactions and mechanisms Implements a user friendly process to achieve the end result of problem solving Covers organic chemistry I and II concepts at the level and depth of a standard ACS organic chemistry curriculum features practice problems and solutions to help master the material including an extensive and comprehensive bank of practice exams with solutions

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*Proceedings of NECC 1979, National Educational Computing Conference* Diana Harris, 1979

**Cumulated Index Medicus**, 1965

**Popular Science**, 2004-12 Popular Science gives our readers the information and tools to improve their technology and their world The core belief that Popular Science and our readers share The future is going to be better and science and technology are the driving forces that will help make it better

*American Universities and Colleges* Praeger Publishers, 2010-04-16 For well over a half century American Universities and Colleges has been the most comprehensive and highly respected directory of four year institutions of higher education in the United States A two volume set that Choice magazine hailed as a most important resource in its November 2006 issue this revised edition features the most up to date statistical data available to guide students in making a smart yet practical decision in choosing the university or college of their dreams In addition the

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Quirk,1994-01-01 This book can be used by students faculty and administrators to enhance the process of medical education Medical students can use it to develop and refine their learning skills assess their individual learning needs identify resources to meet these needs and to prepare themselves for critical learning events and transitions Faculty can use this book to understand the learning process and develop teaching skills thereby providing a framework for critically assessing how they teach Administrators can use it to expose inadequacies in the medical education process and to strengthen the teaching and learning environment Certain chapters of this book could serve as required readings for a medical school course on learning The development and implementation of such a course and the commensurate application of the espoused principles of learning and teaching to the educational process would signal true medical educational reform It would signify that critical attention is being paid to those aspects of learning medicine thought to be crucial reading the literature self directedness critical thinking problem solving lifelong learning and creativity Such a course would be as important as any basic science course or clinical rotation to the training of future physicians who would use these skills throughout their professional lives The book could serve as a foundation for faculty development with medical educators In this regard Part One provides insight into how medical students learn and has significant implications for teaching Parts Two and Three provide practical guidelines for teaching and curriculum development from a learner centered perspective     *El-Hi*

*Textbooks in Print* ,1977     **Psychology of Academic Cheating** Eric M. Anderman,Tamera B. Murdock,2011-04-28 Who cheats and why How do they cheat What are the consequences What are the ways of stopping it before it starts These questions and more are answered in this research based investigation into the nature and circumstances of Academic Cheating Cheating has always been a problem in academic settings and with advances in technology camera cell phones the internet and more pressure than ever for students to test well and get into top rated schools cheating has become epidemic At the same time it has been argued the moral fiber of society as a whole has dampened to find cheating less villainous than it was once regarded Who cheats Why do they cheat and Under what circumstances Psychology of Academic Cheating looks

at personality variables of those likely to cheat but also the circumstances that make one more likely than not to try cheating Research on the motivational aspects of cheating and what research has shown to prevent cheating is discussed across different student populations ages and settings Summarizes 50 years of academic cheating trends in K 12 and postsecondary institutions Examines the methodology of academic cheating including the effect of new technologies Reviews and discusses existing theories and research about the motivation behind academic cheating Resources in Education ,1993-11

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