

Department of Examinations, Sri Lanka

എറണാകുളം സെന്റ് ജോർജ്ജ് ഹയർ സെക്കണ്ടറി സ്കൂൾ (ഗവൺമെൻ്റ്) ലിസ്റ്റംഗലം, 2022(2023)
 കേരള വിദ്യാഭ്യാസ ബോർഡ് പരീക്ഷ (ഉപരി തല) പരീക്ഷ, 2022(2023)
 General Certificate of Education (Adv. Level) Examination, 2022(2023)

Chemical Biology	1
Environmental Chemistry	1
Chemistry	1

02 E I

தமிழ் மொழியை
இருவருள் மனத்திழையாளராம்
Two Inquiries

Instructions:

- Periodic Table is provided.
- This paper consists of 60 pages.
- Answer all the questions.
- Use of calculators is not allowed.
- Write your Index Number in the space provided in the answer sheet.
- Follow the instructions given on the back of the answer sheet carefully.
- In each of the questions 1 to 50, pick one of the alternatives from (1), (2), (3), (4), (5) which is correct or most appropriate and mark your response on the answer sheet with a cross (x) in accordance with the instructions given on the back of the answer sheet.

Universal gas constant $R = 8.314 \text{ J K}^{-1} \text{ mol}^{-1}$

Planck's constant $h = 6.626 \times 10^{-34} \text{ J s}$ Avogadro constant $N_A = 6.022 \times 10^{23} \text{ mol}^{-1}$

Velocity of light $c = 3 \times 10^8 \text{ m s}^{-1}$

- Which of the following electronic transitions belongs to the visible region of the line spectrum of atomic hydrogen? (n = Principal quantum number)
 - $n = 5 \rightarrow n = 3$
 - $n = 4 \rightarrow n = 2$
 - $n = 1 \rightarrow n = 2$
 - $n = 3 \rightarrow n = 1$
 - $n = 2 \rightarrow n = 1$
- Select the incorrect statement.
 - Pauli exclusion principle excludes the possibility of the presence of a third electron in an orbital.
 - In a potassium atom, the number of electrons having quantum numbers n (principal quantum number) = 3 and m_l (magnetic quantum number) = 0 is 4.
 - The effective nuclear charge felt by a valence electron in nitrogen (N), is greater than the effective nuclear charge felt by a valence electron in carbon (C).
 - Among the ions Na^+ , Mg^{2+} , K^+ and Ca^{2+} , the two ions closest to each other in size are K^+ and Mg^{2+} .
 - Electron gain energy of carbon is negative.
- The increasing order of the second ionization energy ($\text{X}^+(\text{g}) \rightarrow \text{X}^{2+}(\text{g}) + e$) of Be, B and O is.
 - $\text{Be} < \text{B} < \text{O}$
 - $\text{Be} < \text{O} < \text{B}$
 - $\text{B} < \text{O} < \text{Be}$
 - $\text{B} < \text{Be} < \text{O}$
 - $\text{O} < \text{Be} < \text{B}$
- The shapes of F_3ClO , FCIO_2 and FCIO_3 are respectively.
 - tetrahedral, trigonal planar and see-saw.
 - square planar, trigonal planar and tetrahedral.
 - see-saw, trigonal pyramidal and square planar.
 - tetrahedral, trigonal pyramidal and see-saw.
 - see-saw, trigonal pyramidal and tetrahedral.

¹ See *note* *supra*.

Gde Chemistry 20memo

Christian G. Meyer



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