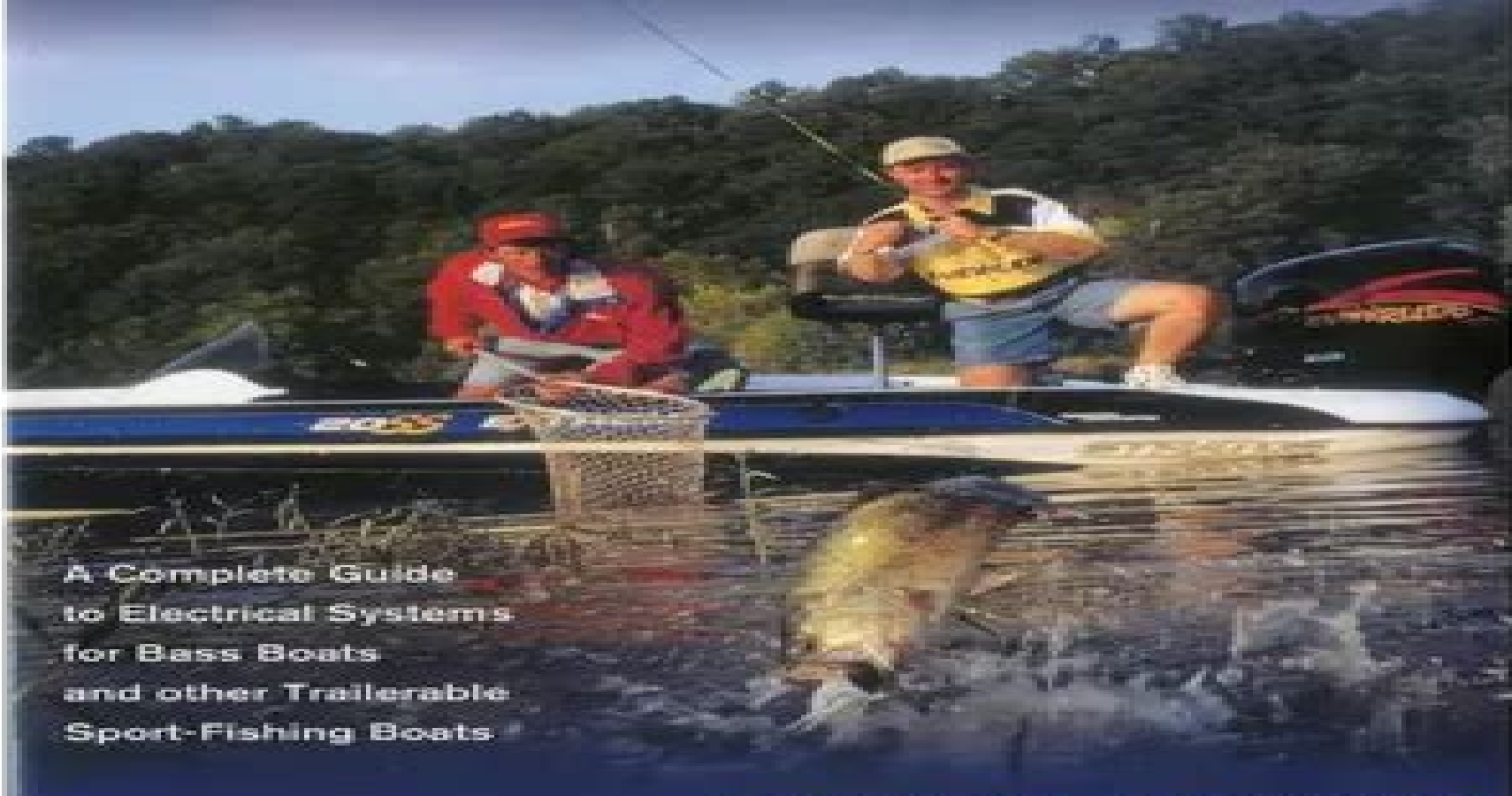


THE FISHERMAN'S ELECTRICAL MANUAL

A photograph of two men on a blue and white fishing boat. The man on the right is wearing a yellow life vest and a cap, holding a fishing rod. The man on the left is wearing a red jacket and a cap, looking at a net. A large fish is jumping out of the water in the foreground. The background shows a forested hillside.

A Complete Guide
to Electrical Systems
for Bass Boats
and other Trailerable
Sport-Fishing Boats

JOHN C. PAYNE

Fishermans Electrical Manual

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copper which is put in the calorimeter are the same the initial temperature of the calorimeter and water is 30 c and that of
hot object is 60 c the specific heat of copper and water are 0 09 cal gm c and 1 cal gm c respectively

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temperature change plugging in the values given in the problem you get $q_{\text{water}} = 418 \text{ J/g} \times 110 \text{ g} \times 26.6 \text{ }^{\circ}\text{C} = 550 \text{ J}$
 $550 \text{ J} = \Delta h \times q_{\text{water}} = 550 \text{ J}$

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web choose 1 answer the final temperature is closer to t_1 than to t_2 a the final temperature is closer to t_1 than to t_2 the final temperature is exactly halfway between t_1 and t_2 b the final temperature is exactly halfway between t_1 and t_2 the final temperature is closer to t_2 than to t_1 c

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