



Name:  e-mail:

**OCR AS Level Chemistry – Paper 1**  
**Unit 1: Atoms, Bonds and Groups**  
 (Each question = 1 mark)

1. Which one of the following statements about the atom is false?
- A A neutron has the same relative mass as a proton.  
 B The number of neutrons in an atom is the same as the number of protons.  
 C The relative mass of an electron is very small compared to that of a proton.  
 D Clouds of negative charge surround the nucleus.

 B


2. The number of neutrons present in the cobalt isotope  $^{60}_{27}\text{Co}$  is

- A 33  
 B 27  
 C 60  
 D 87

 A


3. The analysis of a sample of rubidium using a mass spectrometer showed that two isotopes were present:  $^{85}\text{Rb}$  and  $^{87}\text{Rb}$ . The table below shows the percentage abundance for each isotope:

| isotope                | $^{85}\text{Rb}$ | $^{87}\text{Rb}$ |
|------------------------|------------------|------------------|
| relative isotopic mass | 85.0             | 87.0             |
| % abundance            | 72.1             | 27.9             |

Which one of the following is the relative atomic mass of rubidium to three significant figures?

- A 85.5  
 B 86.5  
 C 85.6  
 D 86.4

 C


# F321 2014 Paper

**S Ashworth**



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