



- 3.5 Substance exists as both a solid and a liquid ✓✓ or... is melting (2)
 3.6 37 °C ✓ (2)
 3.7 80 °C ✓ (1)
 3.8 An indication of the average kinetic energy ✓ of the particles in a substance. ✓ (2)
 3.9 Physical ✓ (1)
 3.10 In the solid state the intermolecular forces are the strongest and the particles cannot move. The heater provides thermal energy to the system, which is translated into kinetic energy (increase in temperature) until the intermolecular forces between the particles can be broken. ✓✓ At this point the thermal energy is transformed into latent energy to break the bonds, so no increase in temperature occurs. ✓✓ (4)

[23]

QUESTION 4

- 4.1
 4.1.1 Homogeneous ✓ (1)
 4.1.2 Homogeneous ✓ (1)
 4.1.3 Pure substance ✓ (1)
 4.2
 4.2.1 Potassium bromide ✓ (2)
 4.2.2 Calcium carbonate ✓ (2)
 4.2.3 Ammonium chloride ✓ (2)
 4.3
 4.3.1 ZnSO_4 ✓✓ (2)
 4.3.2 KMnO_4 ✓✓ (2)
 4.3.3 H_2O ✓✓ (2)

[17]

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Arihant Experts



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Information Engineering and Education Science Dawei Zheng, 2015-04-30 This proceedings volume contains selected papers presented at the 2014 International Conference on Information Engineering and Education Science ICIEES 2014 held June 12 13 in Hong Kong China The objective of ICIEES 2014 was to provide a platform for researchers engineers academics as well as industry professionals from all over the world to

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Annual Report Mpumalanga (South Africa). Department of Education, 2014

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web the lunar new year is an event celebrated by millions of people across the world on the first new moon of their calendar although often referred to as lunar new year in english this is a misnomer as it refers to both celebrations based on a lunar

calendar as well as a lunisolar calendar

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